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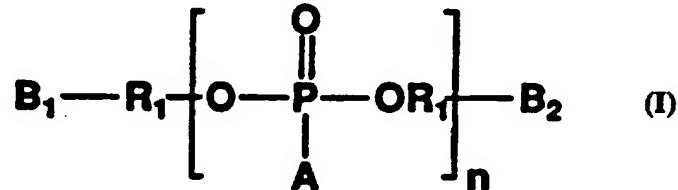
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## (54) Title: NON-NUCLEOTIDE PHOSPHORUS ESTER OLIGOMERS

## (57) Abstract

A phosphorus ester oligomer having structure (I) wherein A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl,



alkyl- or aryl-substituted amino and aminoalkyl; B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonoester; R<sub>1</sub> can be the same or different in each monomeric unit, and in at least one of the non-nucleotide monomeric units, R<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality; (iv) a diol attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label, and n is at least one. Preferred R<sub>1</sub> moieties include condensation products of heterocyclic diols, alicyclic diols, and polycyclic diols. Also the non-nucleotide monomers thereof, combinatorial library mixtures of the oligomers and the use of the oligomers as selective target-binding compounds.

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## NON-NUCLEOTIDE PHOSPHORUS ESTER OLIGOMERS

This invention relates to the field of oligomeric molecules ("oligomers") containing non-nucleotide phosphorus esters, which are useful as mixtures, particularly in the form of combinatorial libraries, to screen for binding activity to biologically significant targets, including protein targets. Esters with high affinity have value as drug or diagnostic candidates, and such combinatorial libraries thereof have value as products used to identify such candidates.

A variety of phosphorus ester oligomers have been reported in the literature. Non-nucleotide phosphorus esters have been incorporated into oligonucleotides as linker groups which connect two separate oligonucleotide sequences. For example, oligonucleotides containing a single non-nucleotide hexaethylene glycol phosphodiester have been described (Durand *et al.*, 1990). The oligonucleotide regions of the molecule were base paired to form a duplex structure, and the non-nucleotide phosphorus ester functioned as a loop to connect the two strands. An aromatic non-nucleotide phosphodiester has also been incorporated into an oligonucleotide as a linker group (Salunkhe *et al.*, 1992), and the non-nucleosides 1,2 dideoxyribose and 1-phenyl-1,2-dideoxyribose have been incorporated into oligonucleotides for enzymatic studies.

Molecules containing multiple non-nucleotide phosphorus esters have also been reported. For example, several 1,3-propanediol phosphodiester groups have been incorporated into oligonucleotides (Richardson and Schepartz, 1991) to act as tethers of varying lengths which connect two separate oligonucleotide sequences. The tethers enabled the oligonucleotide sequences to bind to non-contiguous regions of an RNA target. Diethylene-glycol phosphodiester and deca-ethylene-glycol phosphodiester moieties have also been incorporated into oligonucleotides for similar purposes (Cload and Schepartz, 1991).

Non-nucleotide phosphorus ester oligomers have been used for other functions. For example, oligomeric phosphorus esters possessing substituted alkyl substituents have been synthesized and used as fire retardants (Hardy and Jaffe 1980). Oligomeric phosphorus esters of phenolic compounds have also been described as fireproofing agents (Sase *et al.*, 1988).

In relation to drug discovery, non-nucleotide compounds have been used in combination with nucleotides in a combinatorial manner to search for compounds which

bind to biological targets. Compounds possessing various backbone elements such as ethylene glycol phosphate, hydroxyproline phosphate, or PNA ("peptide nucleic acid") have been used, and a twelve residue pentamer combinatorial library containing substituted ethylene glycol phosphate residues in combination with nucleotide residues was reported (Ecker, 1994). Other groups have described the assembly of combinatorial libraries of oligomers made from a set of monomers consisting of a side chain and a uniform backbone element (Zuckermann et al. 1994). The application of combinatorial libraries to drug discovery has recently been extensively reviewed (Gallop et al., 1994, Gordon et al., 1994).

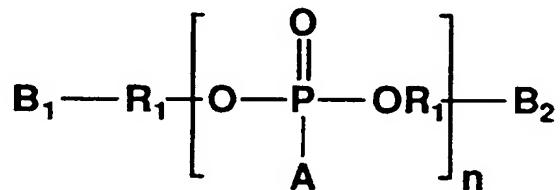
The desirability of introducing conformational constraint into potential drug candidates is a well understood principle in medicinal chemistry. The concept of designing combinatorial libraries of constrained molecules has recently been described, wherein rigid small molecules act as a frame onto which various pendant functions can be attached. Alternatively, the flexibility of members of an oligomer library can be limited by the use of intrinsically constrained amide linking groups, as in the case of PNA or N-substituted glycine libraries (Zuckerman et al., 1994). In the area of antisense oligonucleotides, the phosphodiester linkages have been substituted with acetal groups (Matteucci et al., 1993), and the regular deoxyribose element of the backbone has been replaced with a 5-3-ethano-bridged rigid analog, both with the objective of introducing conformational constraint. In the latter case, the oligonucleotides bound with higher affinity to their cognate sequences than did conventional oligonucleotides (Leumann et al., 1993).

A rapidly growing new industry has developed around the preparation and use of combinatorial libraries of a wide variety of compounds, including polypeptides, small molecules and oligonucleotides. Companies in this industry sell such combinatorial libraries to pharmaceutical, diagnostic and other markets as research reagents suitable for screening and also provide services for companies which do not have the facilities for in house screening of such libraries. The market demand for these combinatorial libraries has been an encouraging demonstration that their utility is well recognized by those skilled in the art.

Combinatorial libraries of oligomers having potential for binding to proteins and other biological targets are valuable products for use in drug screening and other research by pharmaceutical and diagnostic industry members and by investigators in the biological and medical arts. They provide a convenient and rich source of candidates for modulating biologically active agents, particularly proteins, through their binding

potential. Thus these products constitute a pool of readily accessible and sophisticated oligomeric constructs and libraries that satisfies this market and thereby makes structures available to research programs which would not otherwise have access to the broad spectrum of candidates they provide.

One aspect of the invention provides a phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, substituted or unsubstituted alkylamino, substituted or unsubstituted arylamino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonoester;

R<sub>1</sub> can be the same or different in each monomeric unit, and in at least one of the non-nucleotide monomeric units, R<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine- or pyrimidine-substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring substituted-anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label; and

n is at least one.

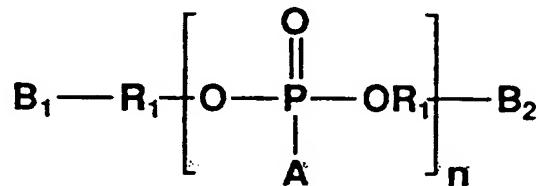
Preferred embodiments of the above aspect include the following:

Preferably, n = 2-20 and more preferably 2-6. Preferably the hydrogen bond donor contains a hydroxyl, amide, imide or thiol group or is a basic compound. The basic compound can contain an amine moiety or a substituted or unsubstituted thiazole.

Preferred hydrogen bond acceptors include an amine or ether moiety. Examples include the 5-substituted 2-hydroxymethyl-3-hydroxy-tetrahydrofurans and bis(hydroxylalkyl)-substituted heterocycles or substituted or unsubstituted theophyllines. Preferred

hydrophobic functionalities are selected from aromatic rings, alkanes, cycloalkanes and aromatic rings fused to alkanes. Particularly preferred hydrophobic functionalities include those wherein R<sub>1</sub> is selected from a substituted alkane, a 3,3-disubstituted 3-amino-1,2-propanediol, a substituted or unsubstituted alicyclic ring wherein the ring size is from 4-12, a 3-substituted indole, a substituted or unsubstituted hydroxyalkyl phenol and an alicyclic dicarboxylic acid. Preferred anionic functionalities include those wherein the anionic functionality is a mono- or dicarboxylic acid moiety. More particularly preferred moieties include tricyclononene dicarboxylic acids and cyclopentane acetic acids. Preferred cationic functionalities include bis(hydroxyalkyl)-substituted nitrogen-containing heterocycles. More particularly preferred cationic functionalities include substituted alkanes and 3,3-disubstituted 3-amino-1,2-propanediols. Another preferred embodiment of the above aspect is an oligomer wherein in at least one of the monomeric units R<sub>1</sub> is a heterocyclic, an alicyclic or a polycyclic ring system, preferably containing an indole, thiazole, imidazole, purine or pyrimidine ring. The alicyclic ring system preferably contains a cyclopentane or cyclooctane ring, and can be substituted with at least one carboxylic acid moiety. The polycyclic ring system preferably contains a bicyclic or tricyclic ring or is a polycyclic arene. The bicyclic ring system is preferably a bicyclic alkane, such as tricyclononene. The polycyclic arene is preferably diphenyl bicyclooctane.

Another aspect of the invention provides a phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

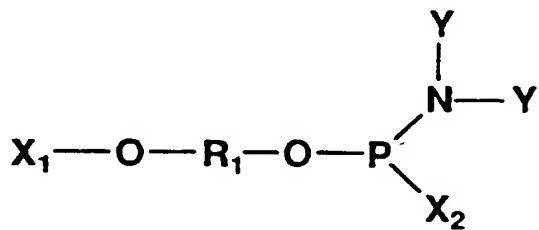
A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, substituted or unsubstituted alkylamino, substituted or unsubstituted arylamino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonooester;

R<sub>1</sub> can be the same or different in each monomeric unit and in at least one of the monomeric units is independently selected from the group consisting of

- (i) an aliphatic acyclic hydrocarbon diol wherein the diol groups are non-vicinal or are substituted;
  - (ii) a purine- or pyrimidine-substituted variant of the diols of (i) or of aliphatic acyclic hydrocarbon vicinal diols;
  - (iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;
  - (iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;
  - (v) a hydroxy or hydroxyalkyl substituted tetrahydrofuran;
  - (vi) an indole-substituted acyclic aliphatic diol;
  - (vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;
  - (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
  - (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline, any of which can further include a detectable label; and
- n is at least 1, preferably 2-20, and more preferably 2-6.

Another aspect of the invention provides a non-nucleotide monomeric unit having the structure:



wherein:

X<sub>1</sub> is a protecting group;

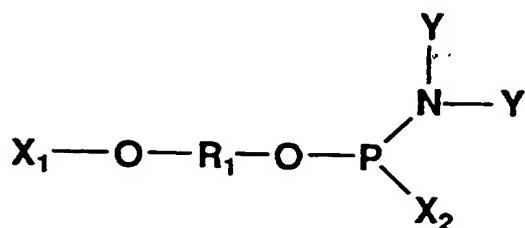
X<sub>2</sub> is a branched or unbranched lower alkyl group or a substituted or unsubstituted alkoxy group;

Y is a branched or unbranched lower alkyl group; and

R<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine- or pyrimidine-substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol

attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol,  
any of which can further include a detectable label.

Another aspect of the invention provides a non-nucleotide monomeric unit having the structure:



wherein

$X_1$  is a protecting group;

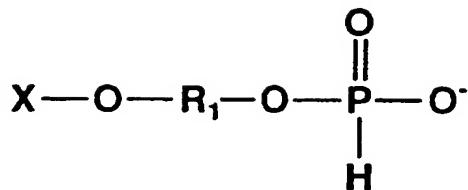
$X_2$  is a branched or unbranched lower alkyl group or a substituted or unsubstituted alkoxy group;

$Y$  is a branched or unbranched lower alkyl group; and

$R_1$  is a condensation product of:

- (i) an aliphatic acyclic diol wherein the diol hydroxyl groups are non-vicinal or are substituted;
  - (ii) a purine- or pyrimidine-substituted variants of the diols of (i) or of aliphatic acyclic hydrocarbon vicinal diols;
  - (iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;
  - (iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;
  - (v) a hydroxy or hydroxyalkyl substituted tetrahydrofuran;
  - (vi) an indole substituted acyclic aliphatic diol;
  - (vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;
  - (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
  - (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline,
- any of which can further include a detectable label.

Another aspect of the invention provides a non-nucleotide monomeric unit having the structure:

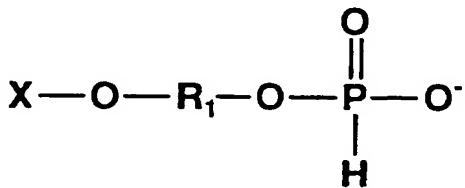


wherein

**X** is a protecting group;

**R**<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine- or pyrimidine-substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring-substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label.

Another aspect of the invention provides a non-nucleotide monomeric unit having the structure:



wherein

**X** is a protecting group;

**R**<sub>1</sub> is a condensation product of: (i) an aliphatic acyclic hydrocarbon diol wherein the diol hydroxyl groups are non-vicinal or are substituted;

(ii) a purine- or pyrimidine-substituted variant of the diols of (i) or of aliphatic acyclic vicinal diols;

(iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;

(iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;

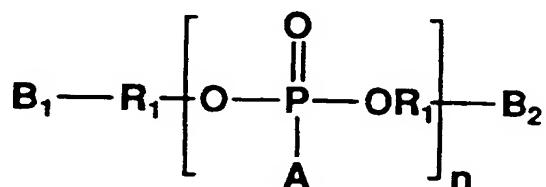
(v) a hydroxy- or hydroxyalkyl-substituted tetrahydrofuran;

(vi) an indole-substituted acyclic aliphatic diol;

(vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;

- (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
- (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline, any of which can further include a detectable label.

Another aspect of the invention provides a phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, alkyl- or aryl-substituted amino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonoester;

R<sub>1</sub> can be the same or different in each monomeric unit, and is selected from the group of a nucleoside moiety and, in at least one monomeric unit, R<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label; and

n is at least one.

Another aspect of the invention provides a combinatorial mixture of the above oligomers. Preferred embodiments of combinatorial mixtures of oligomers include those wherein A is 2-20, and particularly preferred embodiments include those wherein A is 2-6. Preferred embodiments include combinatorial mixtures wherein the oligomers are labelled with <sup>32</sup>phosphorus, <sup>35</sup>sulfur, tritium, fluorescein or biotin.

A large, highly diverse set of structurally unrelated, derivatized monomers is prepared for incorporation into oligomeric libraries. The libraries are synthesized using combinatorial techniques from subsets of available monomers to generate molecular ensembles from which can be selected those with maximum affinity for the target protein. The number of monomers used in the library synthesis is chosen to exploit the full capability of the screening technology used to identify the best ligand.

There is no uniform element in the monomer corresponding to the sugar group of an oligonucleotide, or the amide backbone of peptides, PNA's, hydroxyproline or N-substituted glycine derivatives, or similar types of oligomers. Therefore, there is a high degree of variability in the molecular geometries that can be introduced into the product oligomers.

The nature of the phosphorus linking group in these compounds can be varied at any position to create a greater diversity per unit length of the oligomer than would otherwise be achieved with one type of linkage, such as a phosphodiester or amide. The flexibility of this linkage is also controllable by changing between freely rotatable phosphodiesters, and rotationally constrained phosphoramidates or other esters with bulky groups directly attached to phosphorus. In addition, the gross physical characteristics of the molecules can be varied from anionic to neutral to cationic, with all intermediate cases represented.

Experiments have shown that representative oligomers of this invention are very stable in serum under conditions which rapidly degrade oligonucleotides. Therefore, any lead compound selected from an oligomeric phosphorus ester library is potentially a drug candidate, as compared to oligonucleotide or peptide leads which are readily degraded and therefore most likely require chemical modification to serve as a legitimate drug candidate.

The unique and advantageous aspects of the present invention lie principally in the highly variant set of monomers employed for the synthesis of combinatorial libraries, and the choice of the linking groups in the product oligomers. The monomers fulfill two roles: a structural function whereby the molecular geometry and flexibility of the oligomer are controlled, and secondly, they contain appropriate functional groups to bind to protein or other biological targets. As stated above, the monomers do not contain the sugar or sugar analog element that is found in most other types of oligomers. Instead, any molecule containing two hydroxyl functions may be considered a suitable monomer candidate. By employing such variant compounds to construct oligomers, large

ensembles of molecules with highly variant geometries can be generated. In addition, control can be exercised over the degree of flexibility in these compounds. For example, constraints can be introduced by either using intrinsically inflexible monomers, or monomers in which the two hydroxy groups are vicinal and the resultant steric congestion limits certain bond rotations. Alternatively, it may be desirable to introduce a flexible element into some portion of a molecule, and this could be easily achieved by employing a suitably derivatized  $\alpha,\omega$ -aliphatic diol. The importance of being able to modulate the conformational freedom of potential drug candidates in order to optimize their binding and selectivity profiles is a well understood principal in medicinal chemistry. Also, conformationally-constrained compounds are desirable as lead molecules in any rational drug design program.

In relation to the ability of the monomers to present appropriate binding groups to protein targets, essentially all common organic functional groups, when suitably protected, are compatible with the condensation chemistry that is employed for synthesizing the oligomers. The monomers used for the synthesis of a library directed against a particular biological target are chosen to contribute to the affinity for the target. To this end, the monomers can be subdivided into a range of different categories describing their principal mode of binding, examples being; hydrophobic, hydrogen bond donor, hydrogen bond acceptor, electrostatic cation, and electrostatic anion. Obviously, many monomers contain multiple binding elements and can be assigned to more than one category.

In this invention, each phosphorus linking element connecting the monomer units can be independently modified to both introduce additional binding functionality, and also control the structural flexibility of the oligomers. These linkages can exist either as phosphates, phosphorothioates or phosphoramidates, depending on the oxidation conditions employed for their synthesis. Additional binding groups can be introduced into the oligomers by the oxidation of intermediate H-phosphonate linkages using a diverse range of primary and secondary, simple or highly functionalized amines, an obvious example being amino acids. By using secondary and  $\alpha$ -branched amines, significant rotational constriction of the phosphorus linkage can be achieved, and this constitutes another mechanism for introducing conformational constraint.

To complement the above methods for the introduction of conformational constraints, an extension is the synthesis of libraries of cyclic analogs of oligomeric phosphorus esters. These compounds can be prepared via oxidation reactions of thiol groups to produce cyclic disulfides, such as is described in commonly assigned

copending U.S. patent application Serial No. 08/004,284. This creates structures such as the disulfide-bridged cyclic oligomer as described in Example 44 herein. Other methods for cyclization can be employed, using methods known to those skilled in the art.

A primary utility of oligomeric phosphate ester libraries is the screening of such mixtures for binding activity to particular biological targets, including but not limited to proteins, with the objective of identifying therapeutically important molecules. The structures of the oligomers of the present invention that bind with high affinity to a biological target can be deduced using procedures such as that described in commonly assigned, copending U.S. patent application Serial No. 08/223,519, or by the use of encoded libraries similar to those described in Gallop et al., 1994, or by other methods obvious to those skilled in the art.

Another potential application of this invention is the use of these compounds as glycosaminoglycan mimetics. Glycosaminoglycans are large sulfated oligosaccharides that bind to a range of important regulatory proteins. Work directed at the synthesis of mimetics of these compounds has focused on the preparation of structurally defined heparin oligosaccharides and sulfated dextrans (Lander 1994).

A particular example of this application is in the area of glycosaminoglycan mimetics. Glycosaminoglycans are multiply charged, anionic, sulfated polysaccharides that are typically around 250 saccharide units in length. They are known to modulate a series of biochemical processes by binding to certain polycationic receptor sites on proteins, and play key roles in conditions as diverse as cancer and Alzheimer's disease. The interaction between the protein and the glycosaminoglycan is mostly electrostatic in nature, and consequently the dissociation constant for these interactions is comparatively large, typically in the range  $10^{-5}$  to  $10^{-8}$  M. One approach to the design of mimetics of this class of compound which would have therapeutic application, would require the synthesis of much lower molecular weight anionic molecules, containing certain hydrophobic binding groups. (Lander, 1994). These criteria are met in the oligomeric phosphorus ester libraries of the type discussed in example 43.

In another aspect, the oligomers of the present invention can be used in a method for detecting the presence or absence of a target molecule in a sample, or determining the number of such molecules in a sample. In such methods, the oligomer can be labelled, exposed to target, and the amount of labelled oligomer which is bound to the target is quantified by measurement of the signal derived from the label. The label can be radioactive.

or non-radioactive. Radioactive labels include phosphorus-32, sulfur-35, tritium, and heavy metal isotopes which can be trapped by chelating groups attached to the oligomer. Non-radioactive labels include, but are not limited to, biotin and its analogs, fluorescein, tetramethyl-rhodamine, substrates for enzymes such as alkaline phosphatase or horseradish peroxidase, haptens such as dinitrophenyl or digoxigenin and other labels known to those experienced in the art. Thus, oligomers of the present invention can be used for diagnostics in a manner similar to methods employed for antibody-based diagnostics. The same labelling techniques may also be employed in the protocols for the identification of protein ligands, discussed above.

Figure 1 shows the binding of a combinatorial library of non-nucleotide phosphorus ester oligomers to two protein targets.

Preferred hydrogen bond donors are functionalities containing amine, amide, imide, alcohol and thiol moieties. Examples of such compounds are those wherein R<sub>1</sub> is a 3-substituted dihydroxyalkyl indole (e. g., indolyl-dihydroxypropane), or a bis(hydroxyalkyl)-substituted heterocycle (e. g., 1,2-dihydroxyethylthiazole and pyridoxine) or a 2-amino-1,3-propandiol (e. g.; thiomicamine).

Preferred hydrogen bond acceptor compounds include those containing amine or ether moieties. Examples of such compounds are those wherein R<sub>1</sub> is a purine or pyrimidine substituted 1,2-diol (e. g., theophylline) or a bis (hydroxyalkyl)-substituted heterocycle (e. g., 1,2-dihydroxyethylthiazole and pyridine dimethanol) or a 5-substituted 2-hydroxymethyl 3-hydroxy-tetrahydrofuran (e.g., 1,2 dideoxy-D-ribose).

Preferred hydrophobic groups are alkyl groups and aromatic rings. Examples of such compounds are those wherein R<sub>1</sub> is a substituted 1,3-dihydroxyalkane (e. g. 4-methoxyphenoxy-1,3-propanediol), a substituted 1,2-dihydroxy alkene (e. g.; 1,2-dihydroxy-3-butene), a 3,3-disubstituted 3-amino-1,2-propanediol (e. g., 3-benzyl-3-methylamino-1,2-propanediol and 3,3-diethylamino-1,2-propanediol), a substituted or unsubstituted alicyclic diol wherein the ring size is from 4-12 (e. g. cyclooctanediol and cyclopentanediol), a 3-substituted dihydroxyalkyl indole (e. g., indolyl-dihydroxypropane) or a substituted or unsubstituted hydroxyalkyl phenol (e. g., tetralin and hydroxybenzyl alcohol) or a 1,2-dihydroxy alicyclic, dicarboxylic acid (e. g., tricyclononene dicarboxylic acid).

Preferred electrostatically charged functionalities include anionic functionalities, such as carboxylic, sulfonic and phosphoric acid and tetrazole moieties. Examples of

such compounds are those wherein R<sub>1</sub> is a dihydroxy-substituted carboxylic acid (e. g., cyclopentanediol acetic acid) or a 1,2-dihydroxy alicyclic dicarboxylic acid (e. g., tricyclononene dicarboxylic acid).

Preferred electrostatically charged functionalities also include cationic functionalities. Examples of such compounds are those wherein R<sub>1</sub> is a substituted 1,3-dihydroxyalkane (e. g., 2-amino-1,3-propanediol, 1-phenyl-2-amino-1,3-propanediol and thiomicamine), a 3,3-disubstituted 3-amino-1,2-propanediol (e. g., 3,3-diethylaminopropanediol and 3-benzyl-3-methylamino-1,2-propanediol) or a bis(hydroxyalkyl)-substituted heterocycle (e. g., 1,2-dihydroxyethyl-thiazole and pyridine dimethanol).

Preferred heterocyclic dihydroxy alcohols are those containing an indole, thiazole, imidazole, purine or pyrimidine ring structure. Examples of such compounds are those wherein R<sub>1</sub> is a 3-substituted dihydroxyalkyl indole (e. g., indolyl-dihydroxypropane) or a purine or pyrimidine substituted 1,2-diol (e. g., 2,3-dihydroxypropyl-theophylline) or a bis (hydroxyalkyl)- substituted heterocycle (e. g., thiazole and pyridine dimethanol) or a 5-substituted 2-hydroxymethyl 3-hydroxy-tetrahydrofuran (e.g., 1,2 dideoxy-D-ribose).

Preferred alicyclic dihydroxy alcohols are those containing a cyclopentane or cyclooctane ring structure, including those which are diols and which are substituted with at least one carboxylic acid moiety. Examples of such compounds are those wherein R<sub>1</sub> is a substituted or unsubstituted alicyclic diol wherein the ring size is from 4-12 (e. g. cyclooctanediol and cyclopentanediol) and those wherein R<sub>1</sub> is a dihydroxy-substituted carboxylic acid (e. g., cyclopentanediol acetic acid).

Preferred polycyclic dihydroxy alcohols are those containing a bicyclic or tricyclic ring structure, including alkanes such as a bicycloheptane, alkenes such as tricyclonene diol and polycyclic arenes such as diphenyl bicyclooctane diol. Examples of such compounds are those wherein R<sub>1</sub> is a dihydroxy substituted polycyclic compound (e.g. tricyclononene diol dicarboxylic acid and pinanediol).

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of an aliphatic acyclic hydrocarbon diol wherein the diol hydroxyl groups are non-vicinal or are substituted include those formed from monomers wherein R<sub>1</sub> is 1,3-propanediol or 2-amino-1,3-propanediol.

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of a purine or pyrimidine substituted variant of the diols of (i) or of aliphatic acyclic hydrocarbon vicinal diols include those formed from monomers wherein R<sub>1</sub> is a purine substituted 1,2-diol (e.g. 2,3-dihydroxypropyl theophylline).

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety include those formed from monomers wherein R<sub>1</sub> is 3-diethylamino-1,3-propanediol.

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent include those formed from monomers wherein R<sub>1</sub> is cyclopentanediol, cyclo-octanediol, and those formed from monomers wherein R<sub>1</sub> is a dihydroxy-substituted carboxylic acid (e. g., cyclopentane diol acetic acid) or a 1,2-dihydroxy alicyclic dicarboxylic acid (e. g., tricyclononene diol dicarboxylic acid).

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of a hydroxy or hydroxyalkyl substituted tetrahydrofuran include those formed from monomers wherein R<sub>1</sub> is 1,2-dideoxy-D-ribose.

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of an indole substituted acyclic aliphatic diol include those formed from monomers wherein R<sub>1</sub> is indolyl-dihydroxypropane.

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl include those formed from monomers wherein R<sub>1</sub> is 2,6-bis-hydroxymethyl-pyridine.

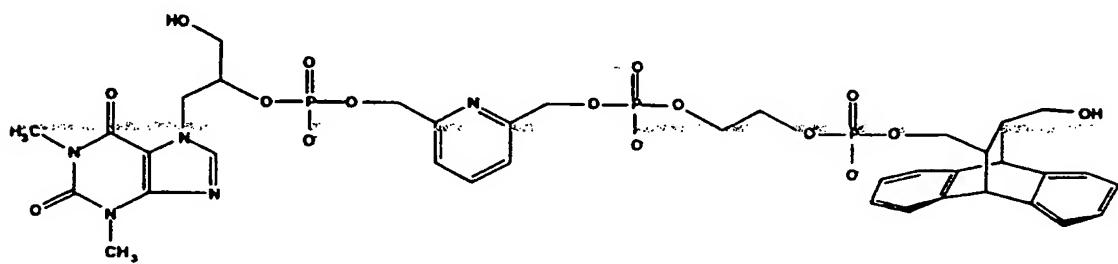
Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl include those formed from monomers wherein R<sub>1</sub> is tetrahydronaphthalene.

Examples of non-nucleotide phosphorus oligomers where R<sub>1</sub> is a condensation product of a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline include those formed from monomers wherein R<sub>1</sub> is 2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole.

The structures of the oligomers of the present invention that bind with high affinity to a protein target can be deduced using procedures including, but not limited to, those described in commonly assigned, copending U.S. patent application Serial No. 08/223,519, or by the use of encoded libraries similar to those described in Gallop *et al.*, 1994.

The libraries are typically synthesized from a diverse pool of about 20 to 30 of such monomer units which are chosen to ensure sufficient diversity with regard to chemical functionality, shape and physical properties, such as charge and hydrophobicity. However, larger or smaller numbers of such monomers can be employed. For the purpose of synthesizing the libraries, these monomers are prepared as 4,4'-dimethoxytrityl (DMT)-protected phosphoramidites and condensed together using known techniques (see Andrus *et al.*, 1988) to produce oligomers with negative charges on the phosphorus backbone. Phosphodiester or phosphorothioate backbones can be produced by these methods. Alternatively, the monomers can be prepared as H-phosphonate derivatives which can be coupled together to produce oligomers using standard procedures (Milligen/Bioscience Inc., Novato CA., H-phosphonate Synthesis Information). The oligomer H-phosphonate intermediates can be subsequently oxidized with iodine and water to produce phosphodiester or with a sulfur reagent to give phosphorothioate linkages.

An example of a simple oligomeric phosphodiester compound of this type which has already been synthesized is as follows:



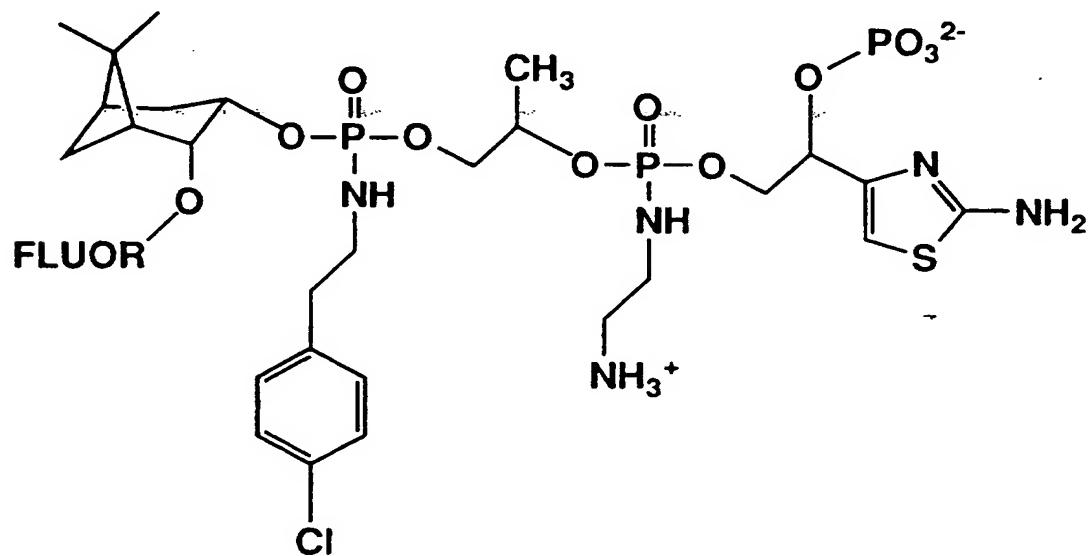
Another aspect of the present invention is the introduction of additional diversity into the backbone of the oligomers by oxidation of the H-phosphonate intermediates with a diverse set of amines to produce phosphoramidate derivatives. The amines can be selected so as to introduce aliphatic groups, aromatic groups, uncharged polar groups such as hydrogen bond donor or acceptor groups, negatively charged groups and/or positively charged groups. Thus the overall structural diversity of the

oligomer can be greatly increased above that which can be achieved when the backbone is comprised exclusively of phosphodiester groups. Examples of amines which can be reacted with oligomer H-phosphonate intermediates to produce oligomer phosphoramidates are as follows:

2-(2-aminoethyl)pyridine; 1-(2-aminoethyl)piperidine; 2-(3, 4-dimethoxyphenyl)ethylamine; 4-(2-aminoethyl)-morpholine; 1-(3-aminopropyl)-4-methylpiperazine; 1-(3-aminopropyl)-2-pyrrolidinone; 1-(3-aminopropyl)imidazole; 1-(2-aminoethyl)pyrrolidine; 3-aminopropionitrile; 2-(2-aminoethyl)-1-methyl-pyrrolidine; 4-fluorophenethylamine; 4-bromophenethylamine; aminomethyl-cyclopropane; 3,3-diphenylpropylamine; formylpiperazine; trifluoromethyl-phenylpiperazine; thiomorpholine; 1-(2-pyridyl)piperazine; homopiperazine; hexamethyleneimine; cis-2,6-dimethylmorpholine; 2,5-dimethylphenylpiperazine; 3,5-dimethylpiperidine; 1-(4-fluorophenyl) piperazine; N-(3, 4-dichlorophenyl)piperazine; 2-(4-chlorophenyl)ethylamine; 4-piperazineacetophenone; 4-piperidinopiperidine; 2-thiophenemethylamine; furfurylamine; heptamethyleneimine; and 1-(4-methoxyphenyl)piperazine.

Other amines can be employed for this purpose as will be evident to those skilled in the art, and amino acid derivatives can also be used in the same way. Thus oligomers can be constructed that can be either anionic, neutral or cationic in character. Using this method, a highly diverse library of molecules with lower molecular weights can be produced.

An example of an oligomeric phosphoramidate of this type which can be synthesized is as follows:



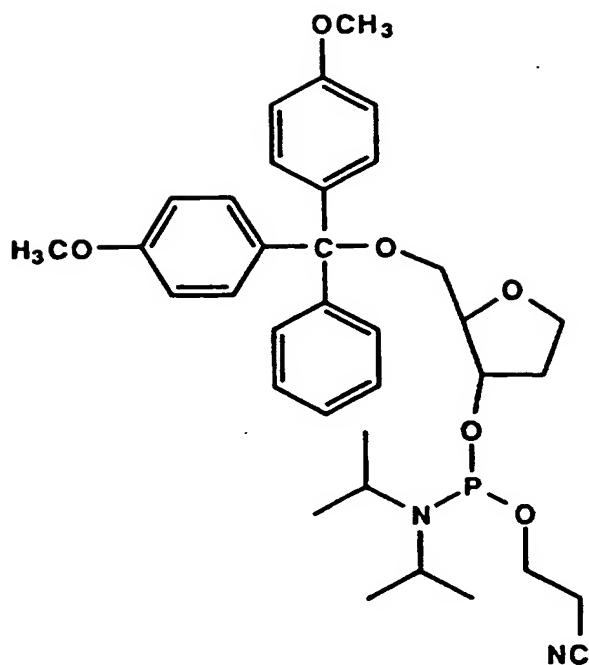
The diversity in this type of oligomer is derived from both the monomer units as described in examples 1-41, as well as the amines which are attached to the phosphorus atoms. The library can be prepared by solid-phase synthesis on a solid support, using a pool and divide strategy employing techniques already reported (Furka *et al.*, 1991).

Known chemical and physical properties of the target can be used to dictate the choice of monomers that are used in the construction of the library. The libraries can be assembled from a monomer subset which is weighted to have a high likelihood of affinity for the binding site on the target as possible, thereby increasing the probability of identifying a suitable ligand. The weighting of the subset can be achieved through the use of molecular modeling techniques,. In the absence of such information, a monomer subset is chosen to be as diverse as is considered desirable. The number of monomers that may be employed for the synthesis of a library is limited by the screening technology used in identifying potential ligands.

#### Example 1

#### Synthesis of 5'-O-(4,4'-dimethoxytrityl)-1,2-dideoxy-D-ribose-3'-O-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

5'-(4,4'-dimethoxytrityl)-1,2-dideoxy-D-ribose (10.7 g) was prepared according to the method of (Grollman *et al*, 1987). This DMT derivative (10.7 g, 25 mmol) was dried over P<sub>2</sub>O<sub>5</sub> under high vacuum, dissolved in dry dichloromethane (60 mL) under nitrogen, then treated with diisopropylethylamine (13.2 g, 102 mmol) and 2-cyanoethoxy-(N,N-diisopropylamino) chlorophosphine (9 g, 38 mmol). After stirring for 2 hours dry methanol (0.5 mL) was added and the reaction mixture was stirred for 20 minutes, poured into 5% aqueous sodium bicarbonate (250 mL), and extracted with ethyl acetate (2<sup>x</sup> 200 mL). The organic layers were washed with saturated aqueous sodium chloride (200 mL), dried overnight over anhydrous sodium sulfate, concentrated to an oil and coevaporated with toluene (20 mL), followed by methanol (20 mL), hexane (20 mL), and dichloromethane (20 mL). The crude product (13.79 g) was purified by Normal Phase High Pressure Liquid Chromatography (NP-HPLC) eluting with a gradient of dichloromethane/hexane (10 to 0%) and then eluting with a gradient of dichloromethane/methanol (0 to 2%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 5'-O-(4, 4'-dimethoxytrityl)-1,2-dideoxy-D-ribose-3'-O-(N,N- diisopropylamino-2-cyanoethyl)-phosphoramidite (13.8 g) as an oil having the following structure:



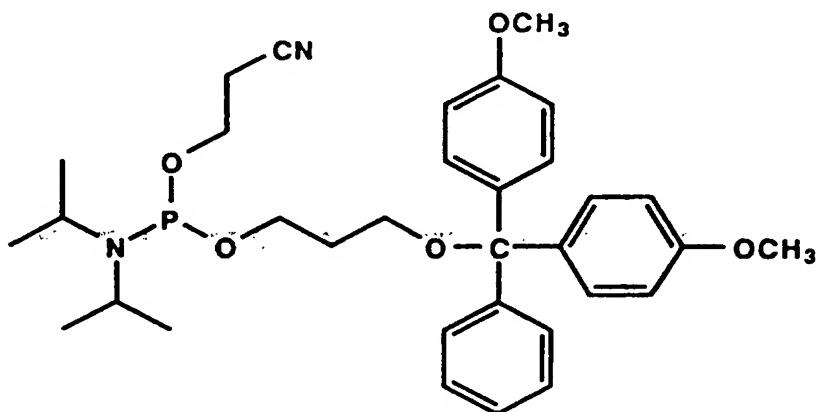
<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.38-6.86 (m, 13H, aromatic), 5.74 (s, 1H), 4.32-4.26 (m, 1H), 3.94-3.44 (m, 3H), 3.72 (s, 6H, OCH<sub>3</sub>), 3.04-2.93 (m, 2H), 2.75-2.71 (m, 2H, CH<sub>2</sub>), 2.64-2.61 (m, 2H, CH<sub>2</sub>), 2.09-2.03 (m, 2H, CH), 1.95-1.91 (m, 1H, H-2"), 1.87-1.83 (m, 1H, H2'), 1.15-0.95 (m, 12H, 4 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 147.5, 147.3.

### Example 2

#### Synthesis of 1-(4,4'-dimethoxytrityl)-propanediol-3-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

1-(4,4'-dimethoxytrityl)-1,3-propanediol (6.8 g) was prepared by the method of Seela and Kaiser (1987). This DMT derivative (6.8 g, 18 mmol) was dissolved in dry dichloromethane (35 mL) under nitrogen and treated with diisopropylethylamine (9.3 g, 72 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (6.4 g, 27 mmol). After stirring for 30 minutes, the mixture was concentrated and 5% aqueous sodium bicarbonate (250 mL) was added. The mixture was extracted with ethyl acetate (3 x 200 mL), the organic layers were washed with saturated aqueous sodium chloride (200 mL), dried over anhydrous sodium sulfate, and concentrated to give the crude product (12 g) as an oil. This was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (50 to 25%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to

give 1-(4, 4'-dimethoxytrityl)-1,3-propanediol-3-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (10.3 g) as a yellow oil having the following structure:

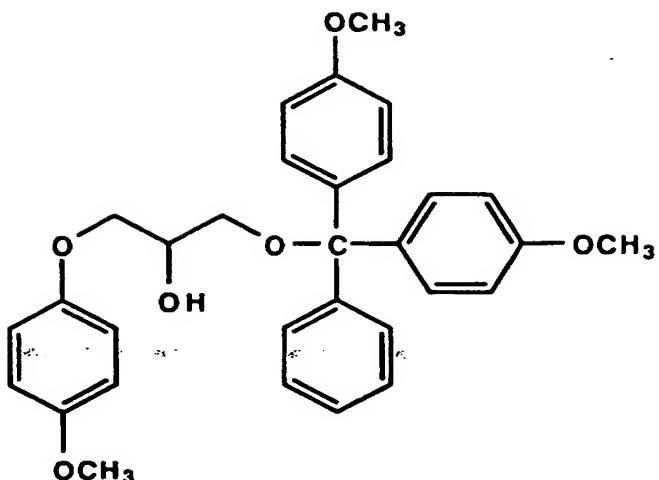


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.37-6.83 (m, 13H, aromatic), 3.72 (s, 6H, OCH<sub>3</sub>), 3.69-3.60 (m, 2H, CH<sub>2</sub>-3), 3.50-3.44 (m, 2H, CH<sub>2</sub>), 3.09-3.01 (m, 2H, CH<sub>2</sub>), 2.68 (t, 2H, J=5.8 Hz, CH<sub>2</sub>-1), 1.84-1.79 (m, 2H, CH<sub>2</sub>-2), 1.10 (d, 6H, J=6.8 Hz, 2 x CH<sub>3</sub>), 1.03 (d, 6H, J = 6.8 Hz, 2 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 146.9.

### Example 3

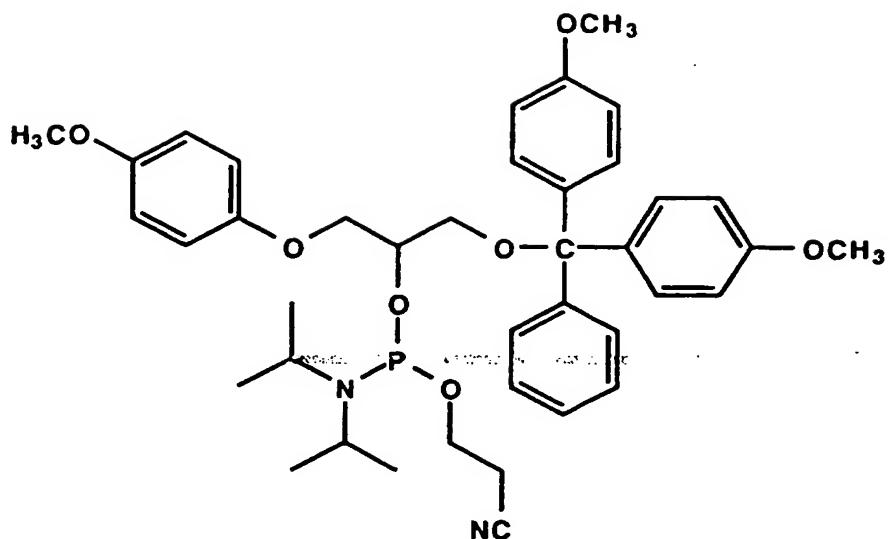
#### Synthesis of 1-(4,4'-dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

3-(4-Methoxyphenoxy)-1,2-propanediol (4 g, 20 mmol) in dry pyridine (50 mL) was treated with DMT chloride (8.2 g, 24 mmol), triethylamine (2.9 g, 28 mmol), and dimethylaminopyridine (120 mg, 1 mmol) and stirred at room temperature for 16 hours. The mixture was poured into 5% aqueous sodium bicarbonate (250 mL), extracted with dichloromethane (3 x 200 mL). The combined organic layers were washed with saturated aqueous sodium chloride (250 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene, methanol, hexane, and dichloromethane to give crude product which was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (90 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol (9.7 g) as a pale yellow oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.39-6.80 (m, 13H, aromatic), 5.09 (d, 1H, J=4.7, OH), 3.96-3.87 (m, 3H, CH<sub>2</sub> & CH), 3.72 (s, 6H, 2 x OCH<sub>3</sub>), 3.68 (s, 3H, OCH<sub>3</sub>), 3.06-3.03 (m, 2H, CH<sub>2</sub>). TLC Rf 0.16 (9:1 dichloromethane/methanol).

This DMT derivative (3.25 g, 6.5 mmol) was dissolved in dry dichloromethane (20 mL) under nitrogen and treated with diisopropylethylamine (2.5 g, 19.5 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (2.3 g, 9.7 mmol). After stirring for 1 hour, the mixture was poured into 5% aqueous sodium bicarbonate (200 mL). The mixture was extracted with ethyl acetate (3 x 200 mL), the organic layers were washed with saturated aqueous sodium chloride (200 mL), dried over anhydrous sodium sulfate, and concentrated to give the crude product (4.8 g) as an oil. This was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (90 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramide (3.4 g, 4.9 mmol) as a colorless oil having the following structure:

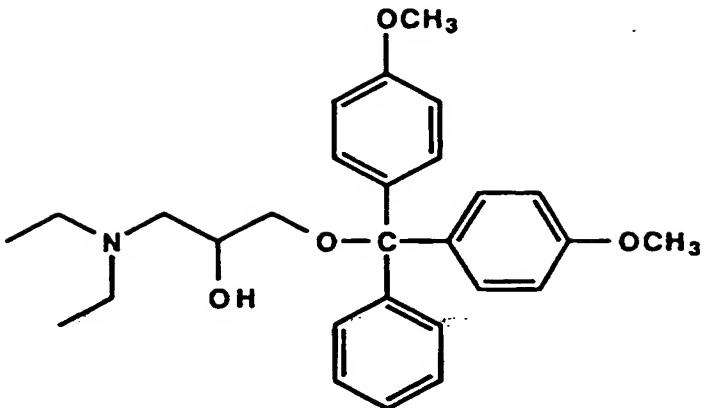


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.40-6.77 (m, 13H, aromatic), 4.21-3.95 (m, 4H,), 3.72 & 3.71 (s, 6H, OCH<sub>3</sub>,2 diastereomers), 3.68 & 3.67 (s, 3H, OCH<sub>3</sub>,2 diastereomers) 3.66-3.46 (m, 4H, CH<sub>2</sub>-3), 3.22-3.09 (m, 3H, CH<sub>2</sub>), 2.72 (t, 1H, J=5.8 Hz, ), 2.63 (t, 1H, J=5.6 Hz), 1.22-0.92 (m, 12H, 4 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 149.6, 149.2.

#### Example 4

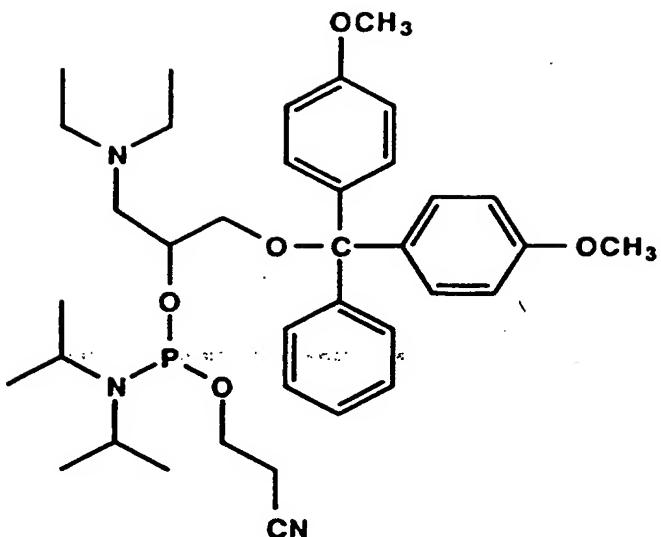
#### Synthesis of 1-(4,4'-dimethoxytrityl)-3-(diethylamino)-1,2-propanediol-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

3-(Diethylamino)-1,2-propanediol (5 g, 34 mmol) in dry pyridine (100 mL) was treated with DMT chloride (13.8 g, 41 mmol), triethylamine (4.8 g, 48 mmol), and dimethylaminopyridine (207 mg, 1.7 mmol) and stirred at room temperature for 4 hours. The mixture was concentrated by rotary evaporation, dissolved in dichloromethane (200 mL) and poured into 5% aqueous sodium bicarbonate (250 mL). The layers were separated and the aqueous layer was extracted with dichloromethane (2 x 200 mL) and the combined organic layers were washed with saturated aqueous sodium chloride (250 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene, methanol, hexane, and dichloromethane to give crude product (20 g) which was purified by column chromatography on silica gel (300 g) eluting with a gradient of dichloromethane/methanol (0 to 4%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-3-(diethylamino)-1,2-propanediol (14.9 g) as a brown oil having the following structure:



$^1\text{H}$  NMR (500 MHz) DMSO  $\delta$  (ppm) 7.42-6.50 (m, 13H, aromatic), 4.47 (br s, 1H, OH), 3.72 (s, 6H, 2 x  $\text{OCH}_3$ ), 3.65 (br s, 1H, CH-2) 2.95-2.88 (m, 2H,  $\text{CH}_2$ -3), 2.40-2.39 (m, 4H, 2 x N- $\text{CH}_2$ ), 2.35-2.28 (m, 2H, CH<sub>2</sub>-1), 0.84 (t, 6H, J = 6.9 Hz, 2 x  $\text{CH}_3$ ). TLC Rf 0.4 (9:1 dichloromethane/methanol).

This DMT derivative (0.5 g, 1.1 mmol) was dissolved in dry dichloromethane (4 mL) under nitrogen and treated with diisopropylethylamine (0.43 g, 3.3 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (0.39 g, 1.7 mmol). After stirring for 2.5 hours, dry methanol (0.5 mL) was added and the mixture was evaporated to dryness, dissolved in 5% aqueous sodium bicarbonate, and extracted with ethyl acetate (3 x 75 mL). The organic layers were washed with saturated aqueous sodium chloride (100 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the mixture evaporated to dryness to give crude 1-(4,4'-dimethoxytrityl)-3-(diethylamino)-1,2-propanediol-2- (N,N-diisopropylamino-2-cyanoethyl)-phosphoramide (0.79 g) as an oil having the following structure:

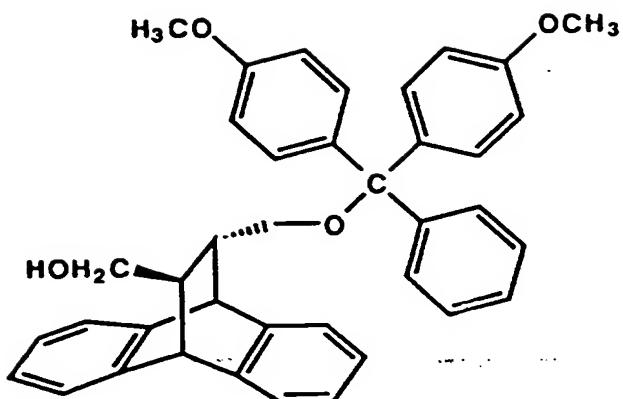


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.48-6.95 (m, 13H, aromatic), 4.57-2.26 (m, 11H,), 3.72 (s, 6H, 2 x OCH<sub>3</sub>), 2.43-2.26 (m, 4H, 2 x CH<sub>2</sub>), 1.25-1.06 (m, 12H, 4 x CH<sub>3</sub>), 0.86-0.73 (m, 6H, 2 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 148.3, 17.3, 16.7, 10.3, 3.7. TLC Rf 0.71 (9:1 dichloromethane/methanol).

### Example 5

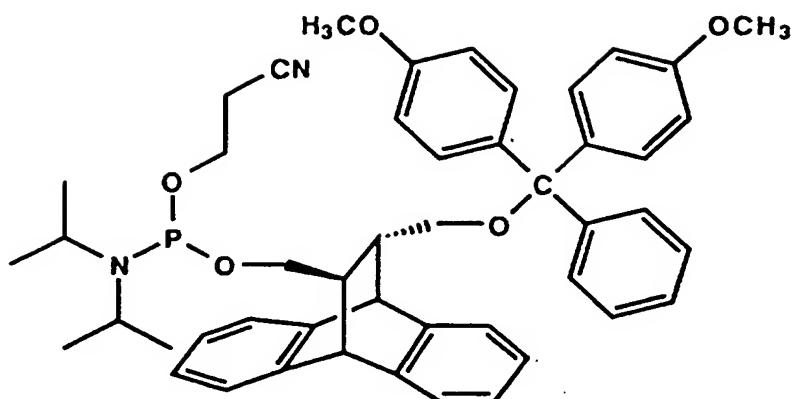
#### Synthesis of 4,4'-dimethoxytrityl-trans-9,10-ethanoanthracene-11,12-dimethanol-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

*trans*-9,10-Ethanoanthracene-11,12-dimethanol (2 g, 7.5 mmol) in dry pyridine (30 mL) was treated with DMT chloride (2.5 g, 7.4 mmol), triethylamine (0.9 g, 8.9 mmol), and dimethylaminopyridine (45 mg, 0.4 mmol) and stirred at room temperature for 1 hour. The mixture was poured into 5% aqueous sodium bicarbonate (200 mL). The mixture was extracted with dichloromethane (4 x 50 mL) and the combined organic layers were washed with saturated aqueous sodium chloride (2 X 50 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene to give crude product (4.9 g) which was purified by column chromatography on silica gel (200 g) eluting with a gradient of dichloromethane/methanol (0 to 3%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 4,4'-dimethoxytrityl-*trans*-9,10-ethanoanthracene-11,12-dim ethanol (2.4 g) as foam having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.36-6.82 (m, 21H, aromatic), 4.59 (d, 1H, J=5.3, OH), 4.40 (d, 1H, J=2.1 Hz, H-9 or 10), 4.26 (d, 1H, J=2.0 Hz, H-10 or 9), 3.713 & 3.707 (s, 6H, 2 x OCH<sub>3</sub>), 3.01 (m, 1H), 2.83 (m, 1H), 2.68 (m, 1H), 2.29 (m, 1H), 1.39 (m, 1H), 1.16 (m, 1H), 0.92 (m, 1H). TLC Rf 0.5 (99:1 dichloromethane/methanol).

This DMT derivative (2.3 g, 4 mmol) was dissolved in dry dichloromethane (15 mL) under nitrogen and treated with diisopropylethylamine (2.1 g, 16 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (1.3 g, 5.6 mmol). After stirring for 1 hour, dry methanol (0.5 mL) was added and the mixture evaporated to dryness. The resulting oil was dissolved in 5% aqueous sodium bicarbonate (300 mL), extracted with ethyl acetate (300 mL, then 100 mL). The organic layers were washed with saturated aqueous sodium chloride (300 mL), dried over anhydrous sodium sulfate, and concentrated to give the crude product as an oil. This was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (80 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 4,4'-dimethoxytrityl-*trans*-9,10-ethanoanthracene-11,12-dim ethanol-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (2 g) as a colorless foam having the following structure:

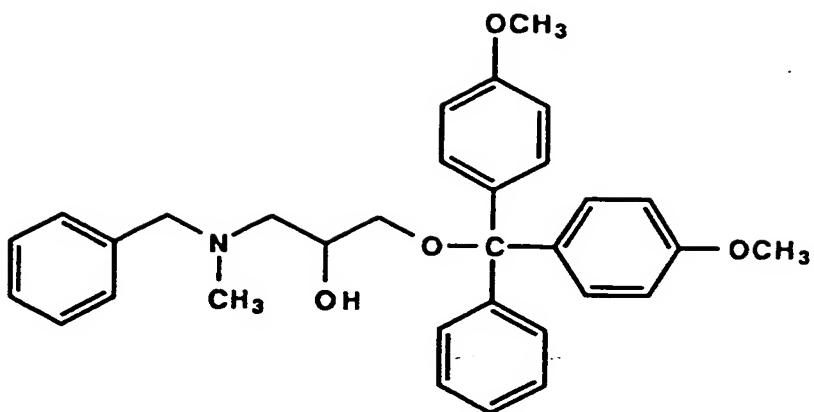


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.36-6.80 (m, 21H, aromatic), 4.41-4.25 (m, 2H, H-9, 10), 3.71 & 3.70 (s, 6H, 2 x OCH<sub>3</sub>), 3.69-3.63 (m, 2H), 3.56-3.48 (m, 3H), 3.27-3.25 (m, 1H), 3.18-3.16 (m, 1H), 2.85-2.80 (m, 2H), 2.76-2.68 (m, 3H), 2.39 (m, 1H), 1.42 (m, 1H), 1.36 (m, 2H), 1.22-0.98 (m, 12H, 4 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 147.1, 146.7.

### Example 6

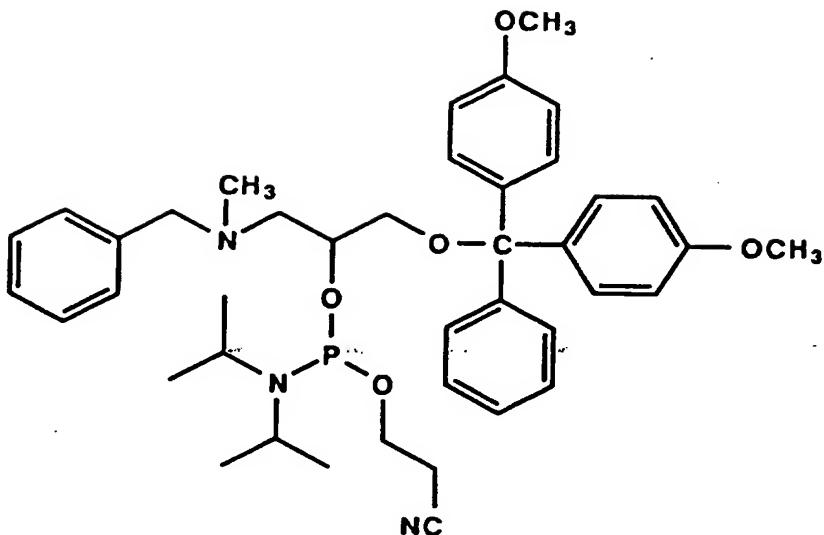
#### Synthesis of 1-(4,4'-dimethoxytrityl)-3-(N-benzyl-N-methylamino)-1,2-propanediol-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

3-(N-Benzyl-N-methylamino)-1,2-propanediol (5 g, 26 mmol) in dry pyridine (50 mL) was treated with DMT chloride (10.4 g, 31 mmol), triethylamine (3.6 g, 36 mmol), and dimethylaminopyridine (160 mg, 1.3 mmol) and stirred at room temperature for 2 hours. The mixture was poured into 5% aqueous sodium bicarbonate (300 mL). The layers were separated, the aqueous layer was extracted with dichloromethane (3 x 200 mL), the combined organic layers were washed with saturated aqueous sodium chloride (300 mL), and dried over anhydrous sodium sulfate. The solid was filtered off, the filtrate was evaporated to dryness, and coevaporated with toluene and dichloromethane to give crude product (14 g) which was purified by column chromatography on silica gel (300 g) eluting with a gradient of dichloromethane/methanol (0 to 4%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-3-(N-benzyl-N-methyl)-1,2- propanediol (12.1 g) as a brown oil having the following structure: . . .



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.40-6.85 (m, 18H, aromatic), 4.64 (d, 1H, J=4.8 Hz, OH), 3.81 (q, 1H, J=5.3 Hz, CH-O), 3.71 & 3.70 (s, 6H, OCH<sub>3</sub>, 2 diastereomers), 3.72(s, 6H, OCH<sub>3</sub>), 3.46 (AB quartet, 1H, J<sub>AB</sub>=13.3 Hz, benzylic H), 3.38 (AB quartet, 1H, J<sub>AB</sub>=13.3 Hz, benzylic H), 2.94 (m, 2H, CH<sub>2</sub>-N), 2.36 (m, 2H, CH<sub>2</sub>-O), 2.08 (s, 3H, N-CH<sub>3</sub>). TLC Rf 0.6 (9:1 dichloromethane/methanol).

This DMT derivative (2.4 g, 4.8 mmol) was dissolved in dry acetonitrile (5 mL) under nitrogen and treated with diisopropylamine (0.73 g, 7.2 mmol), tetrazole (340 mg, 4.8 mmol), and 2-cyanoethoxy-(N,N,N,N-tetraisopropylamino)phosphine (3.63 g, 12.5 mmol). After stirring for 16 hours at room temperature, the mixture was evaporated to dryness, dissolved in ethyl acetate (200 mL) and washed with 5% aqueous sodium bicarbonate (200 mL). The layers were separated and the organic layer was washed with saturated aqueous sodium chloride (200 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the mixture evaporated to dryness to give the crude product (5.4 g) as a yellow oil which was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (20 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-3-(N-benzyl-N-methylamino)-1,2-propanediol-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (3.3 g) as an oil having the following structure:

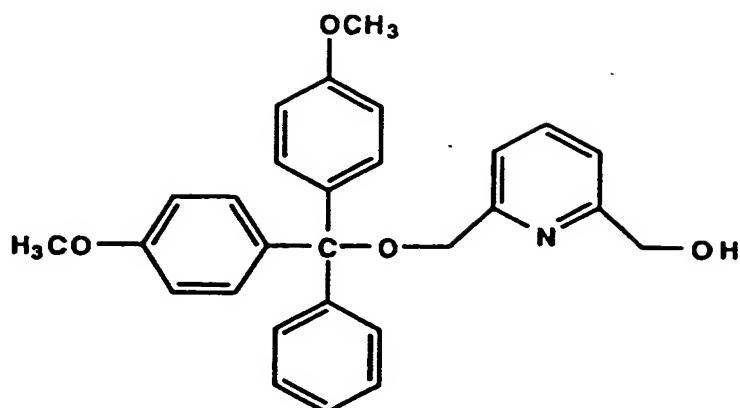


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.42-6.77 (m, 18H, aromatic), 4.08-3.06, (m, 11H), 3.73 (s, 6H, 2 x OCH<sub>3</sub>), 2.78-2.42 (m, 4H, 2 x CH<sub>2</sub>), 1.25-1.06 (m, 12H, 4 x CH<sub>3</sub>), 0.86-0.73 (m, 6H, 2 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 148.6, 148.5, 140.2, 139.5, 124.1 TLC Rf 0.92, 0.82 (9:1 dichloromethane/methanol).

### Example 7

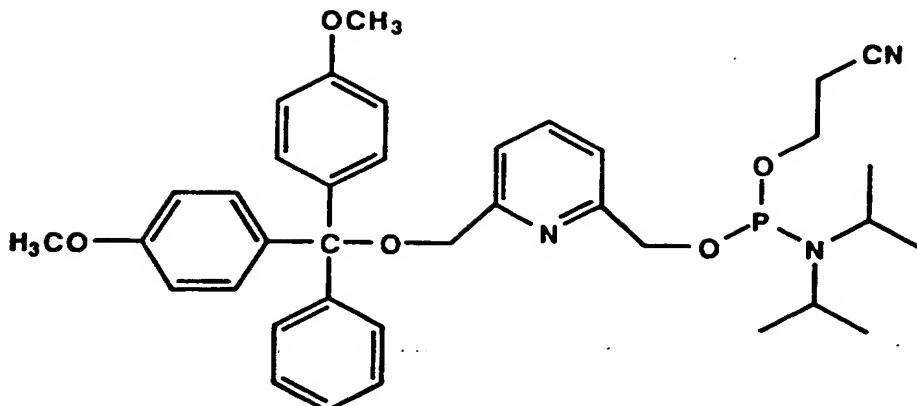
#### Synthesis of 2-(4,4'-dimethoxytrityl)-2,6-bis-hydroxymethylpyridine-6-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

2,6-bis-hydroxymethylpyridine (2.1 g, 15 mmol) in dry pyridine (50 mL) was treated with DMT chloride (1 g, 3 mmol), triethylamine (0.37 g, 3.6 mmol), and dimethylaminopyridine (20 mg, 0.15 mmol) and stirred at room temperature for 16 hours. The mixture concentrated by rotary evaporation and poured into 5% aqueous sodium bicarbonate (200 mL), then extracted with dichloromethane (3 x 200 mL) and the combined organic layers were washed with saturated sodium chloride solution (200 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene and dichloromethane to give crude product which was purified by column chromatography on silica gel (300-g) eluting with a gradient of dichloromethane/methanol (0 to 4%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 2-(4,4'-dimethoxytrityl)-2,6-bis-hydroxymethylpyridine (3.8 g) as an oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.87-6.90 (m, 16H, aromatic), 5.29 (s, 1H, OH), 4.44 (s, 2H, CH<sub>2</sub>), 4.06 (s, 2H, CH<sub>2</sub>), 3.73 (s, 6H, 2 x OCH<sub>3</sub>). TLC Rf 0.68 (9:1 dichloromethane/methanol).

This DMT derivative (3.3 g, 5.2 mmol) was dissolved in dry acetonitrile (65 mL) under nitrogen and treated with tetrazole (0.33 g, 4.6 mmol), diisopropylamine (0.68 g, 6.7 mmol), and 2-cyanoethoxy-(N,N,N,N-tetraisopropylamino)phosphine (3.3 g, 10.9 mmol). After stirring for 30 minutes, additional diisopropylamine (0.7 g, 6.7 mmol), and 2-cyanoethoxy-(N,N,N,N-tetraisopropylamino)-phosphine (1.8 g, 6.2 mmol) was added and the reaction mixture was stirred for 1 hour. The mixture was poured into 5% aqueous sodium bicarbonate (250 mL), extracted with dichloromethane (2 x 100 mL, 2 x 200 mL, 6 x 50 mL), the organic layers were washed with saturated aqueous sodium chloride (250 mL), dried over anhydrous sodium sulfate, and concentrated to give the crude product (7 g) as an oil. This was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (10 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 2-(4,4'-dimethoxytrityl)-2,6-bis-hydroxymethylpyridine-6-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (5.4 g) as a pale yellow oil having the following structure:

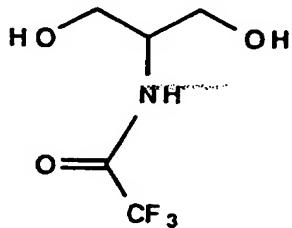


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.91-6.20 (m, 16H, aromatic), 4.64 (m, 2H, CH<sub>2</sub>-6), 4.09 (s, 2H, CH<sub>2</sub>-2), 3.73 (s, 6H, 2 x OCH<sub>3</sub>), 3.82-3.46 (m, 6H, CH<sub>2</sub> & CH), 1.24-1.10 (m, 12H, 4 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 148.6, 139.1, 123.6. TLC Rf 0.77, 0.68 (1:1 ethyl acetate/dichloromethane).

#### Example 8

#### Synthesis of 1-(4,4'-dimethoxytrityl)-2-N-trifluoroacetyl-2-amino-1,3-propanediol-3-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

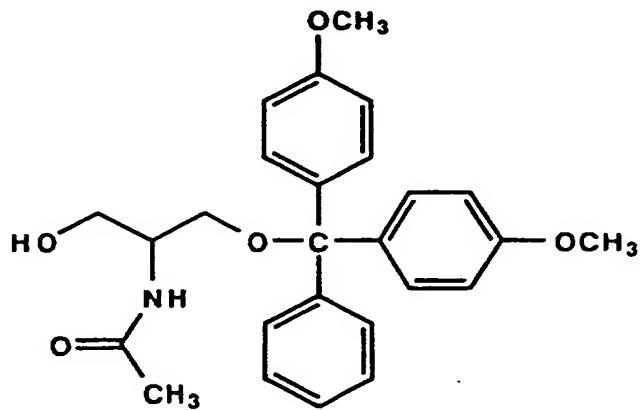
Serinol (1 g, 11 mmol) in dry dichloromethane (10 mL) and dry pyridine (2.7 mL) was cooled to -78 °C. on a dry ice/acetone bath and treated with trifluoroacetic anhydride for 1 hour. The reaction mixture was evaporated to dryness and coevaporated with toluene, methanol, hexane, and dichloromethane to give crude 2- N-trifluoroacetamido-serinol (4.2 g) as an oil with the following structure:



An analytical sample (100 mg) was purified by column chromatography on neutral alumina eluting with isocratic methanol.

<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 4.69 (t, 2H, J=5.7 Hz, OH), 3.80 (m, 1H, CH), 3.46 (m, 2H, 2 x CH<sub>2</sub>) 3.17 (d, 1H, J=5.2 Hz, NH). TLC Rf 0.44 (7:3 dichloromethane/ methanol) visualized with ninhydrin spray.

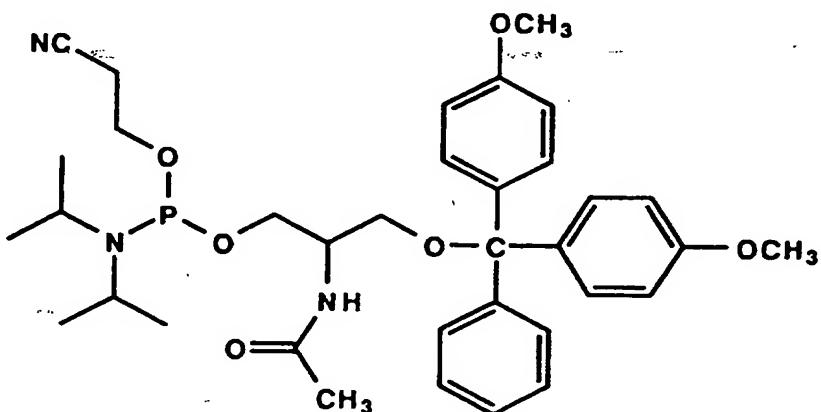
This material (10.3 g, 55 mmol) in dry pyridine (200 mL) was treated with DMT chloride (16.7 g, 49 mmol), triethylamine (7.8 g, 77 mmol), and dimethylaminopyridine (335 mg, 2.7 mmol). Additional DMT chloride (3.35, 9.9 mmol), triethylamine (1.8 g, 18 mmol), and dimethylaminopyridine (70 mg, 0.6 mmol) was added after 18 hours and at 26 hours, and the reaction mixture was stirred at room temperature for a total of 90 hours. The mixture was poured into 5% aqueous sodium bicarbonate (400 mL) and extracted with dichloromethane (500 mL) and the combined organic layers were washed with saturated aqueous sodium chloride (300 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene, methanol, hexane, and dichloromethane to give crude product (25 g) which was purified by column chromatography on silica gel (280 g) eluting with a gradient of dichloromethane/methanol (0 to 3%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-2-N-trifluoroacetamido-serinol (13.5 g) as an having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 9.16 (d, 1H, J= 8.4 Hz, NH), 7.77-6.82 (m, 13H, aromatic), 4.70 (t, 1H, J= 5.5 Hz, OH), 4.08 (m, 1H, CH), 3.73 (s, 6H, 2 x OCH<sub>3</sub>), 3.14-3.11 (m, 2H, CH<sub>2</sub>), 3.01-2.98 (m, 2H, CH<sub>2</sub>). TLC Rf 0.37 (99:1 dichloromethane/ methanol).

This DMT derivative (4.4 g, 9 mmol) was dissolved in dry dichloromethane (20 mL) under nitrogen and treated with diisopropylethylamine (3.5 g, 27 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (3.2 g, 13.5 mmol). After stirring for 45 minutes the mixture was poured into 5% aqueous sodium bicarbonate (200 mL), and extracted with ethyl acetate (200 mL). The organic layers were washed with saturated aqueous sodium chloride (200 mL) and dried over anhydrous sodium sulfate.

The solid was filtered off and the mixture evaporated to dryness to give crude product (7.2 g) which was purified by NP-HPLC using a gradient of dichloromethane/hexane (40 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-2-N-trifluoroacetamido-serinol-3-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (5.2 g) as an oil having the following structure:

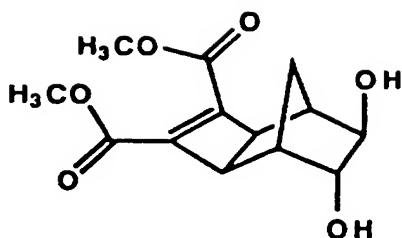


<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 9.40 (m, 1H, NH), 7.36-6.83 (m, 13H, aromatic), 4.22 (m, 1H, CH), 3.72 (s, 6H, 2 x OCH<sub>3</sub>), 3.67-2.62 (m, 10H, 4 x CH<sub>2</sub>, 2 x CH), 1.24-0.85 (m, 12H, 4 x CH<sub>3</sub>), 0.86-0.73 (m, 6H, 2 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 150.3. TLC Rf 0.71 (9:1 trichloroethane/ethyl acetate)

#### Example 9

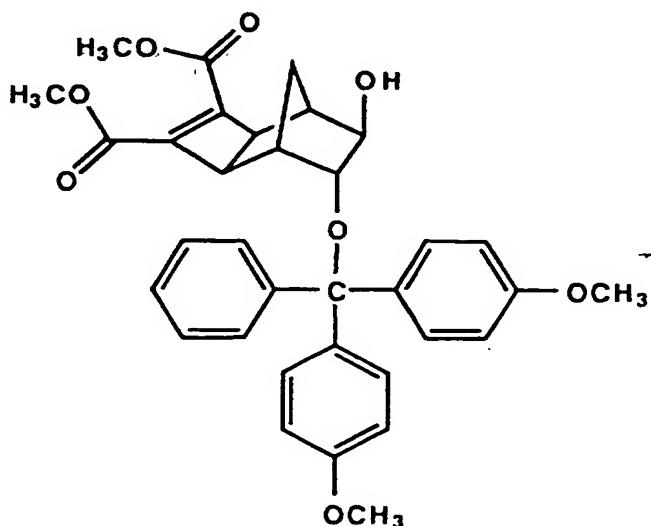
#### Synthesis of 7-(4,4'-dimethoxytrityl)-trans-7,8-dihydroxy-dimethyl-exo-tricyclononene[4.2.1.0<sup>2,5</sup>]nona-3-ene-3,4-dicarboxylate-8-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

Dimethyl<sup>2</sup>exo-tricyclo[4.2.1.0<sup>2,5</sup>]-3,7-diene-3,4-dicarboxylate (1 g, 4.3 mmol) was added dropwise to a mixture of 88% formic acid (2.6 mL, 59 mmol) and 30% hydrogen peroxide (0.6 mL, 6 mmol) that had been cooled on an ice bath. The ice bath was removed and the mixture was stirred for 18 hours. The mixture was evaporated to dryness, co-evaporated with methanol, toluene, methanol, and dichloromethane. This residue was dissolved in methanol (10 mL), Amberlite® IR- 120 strongly acidic ion-exchange resin (1 g) was added, and the mixture was refluxed for 2 hours. The solid was filtered off and the filtrate evaporated to give crude *trans*-dimethyl<sup>2</sup>exo-tricyclo[4.2.1.0<sup>2,5</sup>]-3-ene-7,8-diol-3,4 -dicarboxylate (1.4 g) as a brown oil having the following structure:



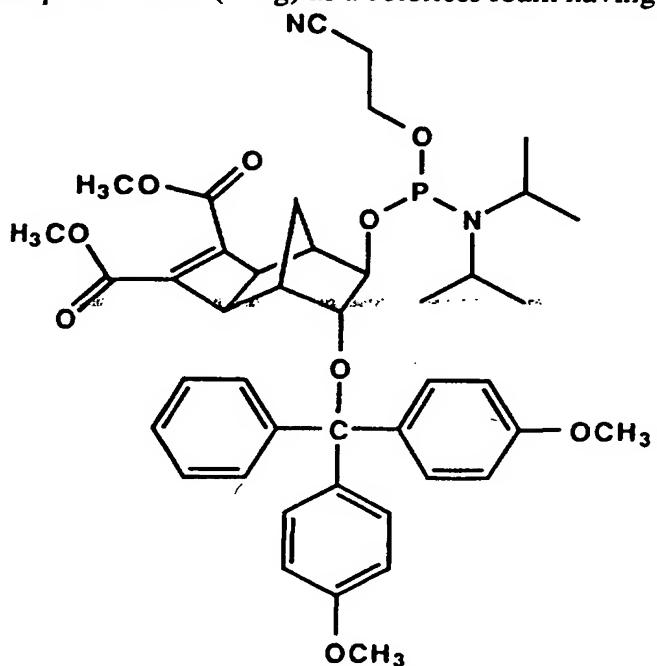
<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 5.06 (d, 1H, J=5.6 Hz, OH), 4.39 (d, 1H, J=7.0 Hz, OH), 3.93 (m, 1H, CHO), 3.714 (s, 3H, OCH<sub>3</sub>); 3.710 (s, 3H, OCH<sub>3</sub>), 3.65 (m, 1H, CHO), 3.02 (m, 1H, CH-1 or 6), 2.92 (m, 1H, CH-6 or 1), 2.37 (m, 1H, CH-2 or 5), 2.21 (m, 1H, CH-5 or 2), 1.65 (m, 2H, CH<sub>2</sub>).

This material (1.4 g, 4.3 mmol) in dry pyridine (20 mL) was treated with DMT chloride (1.3 g, 3.9 mmol), triethylamine (290 mg, 5.2 mmol), and dimethylamino-pyridine (2 mg, 0.02 mmol) and stirred at room temperature for 4 hours. The mixture was poured into 5% aqueous sodium bicarbonate (100 mL). The layers were separated and the aqueous layer was extracted with dichloromethane (6 x 30 mL) and the combined organic layers were washed with saturated aqueous sodium chloride (2 x 30 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene, methanol, and dichloromethane to give crude product (2.4 g) which was purified by column chromatography on silica gel (100 g) eluting with a gradient of dichloromethane/methanol (0 to 1%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 7-(4,4'-dimethoxytrityl)-*trans*-7,8-dihydroxy-dimethyl-*exo*- tricyclononene[4.2.1.0<sup>2,5</sup>]nona-3-ene-3,4-dicarboxylate (1.2 g) as a yellow foam having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.36-6.82 (m, 13H, aromatic), 4.65 (d, 1H, J=5.0 Hz, OH), 3.76 (m, 1H, OCH), 3.724 & 3.721 (s, 6H, 2 x OCH<sub>3</sub>, 2 diastereomers), 3.63 (s, 3H, OCH<sub>3</sub>), 3.47 (m, 1H, OCH), 3.38 (s, 3H, OCH<sub>3</sub>), 2.90 (m, 1H, CH-1 or 6) 2.77 (m, 1H, CH-6 or 1), 2.09 (m, 1H, CH-2 or 5), 2.02 (m, 1H, CH-5 or 2), 1.44 (m, 1H, CH<sub>2</sub>), 0.82 (m, 1H, CH<sub>2</sub>). TLC Rf 0.28 (dichloromethane).

This DMT derivative (1.2 g, 2.1 mmol) was dissolved in dry acetonitrile (20 mL) under nitrogen and treated with tetrazole (150 mg, 2.1 mmol), diisopropylamine (0.32 g, 3.15 mmol), and 2-cyanoethoxy-(N,N,N,N-tetraisopropylamino)phosphine (1.6 g, 5.3 mmol). After stirring for 21 hours the mixture was evaporated to dryness, dissolved in 5% aqueous sodium bicarbonate (125 mL), and extracted with dichloromethane (2 x 100 mL). The organic layers were washed with water (125 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the mixture evaporated to dryness to give crude product (2.6 g) which was purified by NP-HPLC eluting with a gradient of dichloromethane/hexane (60 to 0%). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 7-(4,4'-dimethoxytrityl)- trans-7,8-dihydroxy-dimethyl-*exo*-tricyclo[4.2.1.0<sup>2,5</sup>]nona-3-ene-3,4- dicarboxylate-8-(N,N- diisopropylamino-2-cyanoethyl)-phosphoramidite (1.4 g) as a colorless foam having the following structure:



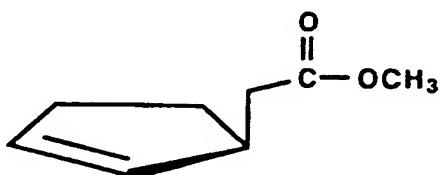
<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.38-6.80 (m, 13H, aromatic), 3.92 (m, 1H, OCH), 3.82 (m, 1H, OCH), 3.72 & 3.71 (s, 6H, 2 x OCH<sub>3</sub> for 2 diastereomers), 3.63

& 3.629 (s, 3H, OCH<sub>3</sub> for 2 diastereomers), 3.353 & 3.347 (s, 3H, OCH<sub>3</sub> for 2 diastereomers), 2.96 (m, 1H, CH-1 or 6), 2.82 (m, 1H, CH-6 or 1), 2.72 (m, 4H, 2 x CH<sub>2</sub>), 2.26 (m, 1H, H-2 or 5), 2.18 (m, 1H, CH-5 or 2), 1.39 (m, 1H, CH<sub>2</sub>), 1.19-1.05 (m, 12H, 4 x CH<sub>3</sub>), 0.80 (m, 1H, CH<sub>2</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 146.8, 146.4, 123.7. TLC Rf 0.56 (99:1 dichloromethane/methanol).

#### Example 10

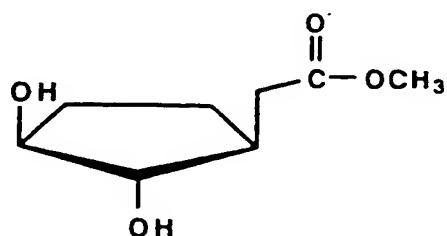
##### Synthesis of 1-(4,4'-dimethoxytrityl)-trans-1,2-cyclopentanediol-3-methylacetate-2-(N,N-diisopropylamino-2-cyanoethyl) phosphoramidite

2-Cyclopentene-1-acetic acid (2.35g, 18.7 mmol) in diethyl ether (5 mL) was cooled on an ice/salt bath and treated with diazomethane (50 mL, 0.46 M) at 0 °C. The mixture was concentrated to dryness to give 2-cyclopentene-1-methylacetate (3.1 g) as a colorless oil with the following structure:



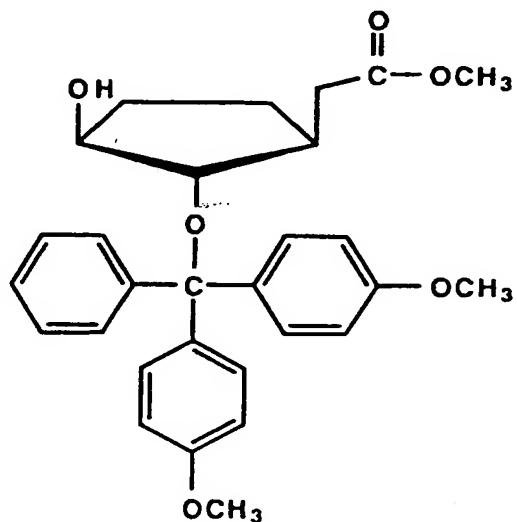
<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 5.74 (m, 1H, vinylic CH), 5.65 (m, 1H, vinylic CH), 3.58 (s, 3H, OCH<sub>3</sub>), 2.95 (m, 1H, CH), 2.39-2.18 (m, 4H, CH<sub>2</sub>), 1.98 (m, 1H, CH<sub>2</sub>), 1.38 (m, 1H, CH<sub>2</sub>), 1.65 (m, 2H, CH<sub>2</sub>). TLC Rf 0.89 (9:1 dichloromethane/methanol), visualized with ammonium molybdate/sulfuric acid.

This material (1.5 g) was added dropwise to a mixture of 96% formic acid (5.6 mL, 150 mmol) and 30% hydrogen peroxide (0.5 mL, 15 mmol) that had been cooled on an ice/salt bath. The ice bath was removed and the mixture was stirred for 16 hours. The mixture was evaporated to dryness, co-evaporated with toluene, methanol, and hexane. This residue was dissolved in methanol (30 mL), Amberlite® IR-120 strongly acidic ion-exchange resin (3 g) was added, and the mixture was refluxed for 2 hours. The solid was filtered off and the filtrate evaporated to give crude trans-1,2-cyclopentanediol-3-methylacetate (1 g) as an oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 4.90 (br s, 2H, OH), 4.57 (m, 1H), 4.04 (m, 1H), 3.57 (s, 3H, OCH<sub>3</sub>), 2.91 (m, 1H), 2.78 (m, 1H) 2.22 (m, 1H), 2.03 (m, 1H, CH<sub>2</sub>), 1.68 (m, 1H, CH<sub>2</sub>), 1.53 (m, 1H, CH<sub>2</sub>), 1.42 (m, 1H, CH<sub>2</sub>).

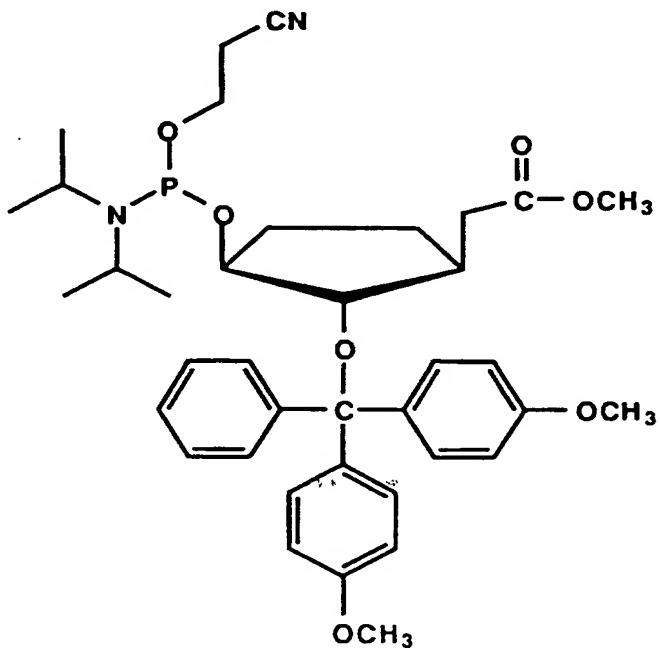
This material (1 g, 6 mmol) in dry pyridine (20 mL) was treated with DMT chloride (2.4 g, 7.2 mmol), triethylamine (0.85 g, 8.4 mmol), and dimethylaminopyridine (37 mg, 0.3 mmol) and stirred at room temperature for 4 hours. The mixture was concentrated by rotary evaporation, dissolved in dichloromethane (200 mL) and poured into 5% aqueous sodium bicarbonate (200 mL). The layers were separated and the aqueous layer was extracted with dichloromethane (2 x 200 mL) and the combined organic layers were washed with saturated aqueous sodium chloride (250 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene, methanol, hexane, and dichloromethane to give crude product (3.3 g) which was purified by column chromatography on silica gel (150 g) eluting with a gradient of dichloromethane/methanol (0 to 2%). The purities of the fractions were monitored by TLC and fractions containing the desired material (520 mg) were combined, evaporated to dryness and repurified by NP-HPLC eluting with isocratic dichloromethane. The purities of the fractions were monitored by TLC and fractions containing the pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-*trans*-1,2-cyclopentanediol-3-methyl acetate (260 mg) as a pale yellow oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.47-6.83 (m, 13H, aromatic), 4.65 (d, 1H, J=5.5 Hz, OH), 3.73 (s, 6H, 2 x OCH<sub>3</sub>), 3.58 (m, 2H), 3.55 (s, 3H, OCH<sub>3</sub>), 3.49 (m,

2H), 2.54 (m, 2H) 2.20 (m, 2H), 1.81 (m, 2H), 1.50 (m, 2H), 1.14 (m, 2H), 0.87 (m, 2H), 0.77 (m, 2H). TLC R<sub>f</sub> 0.32 (9:1 trichloroethane/ethylacetate).

This DMT derivative (260 mg, 0.6 mmol) was dissolved in dry dichloromethane (5 mL) under nitrogen and treated with diisopropylethylamine (0.23 g, 1.8 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (0.21 g, 0.9 mmol). After stirring for 15 minutes the mixture was evaporated to dryness, dissolved in ethylacetate (75 mL), washed with 5% aqueous sodium bicarbonate (100 mL). The aqueous layer was extracted with ethylacetate (2 x 75 mL). The organic layers were washed with saturated aqueous sodium chloride (100 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the mixture evaporated to dryness to give crude product (540 mg) which was purified by NP-HPLC eluting with isocratic dichloromethane/hexane (6:4). The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 1-(4,4'-dimethoxytrityl)-trans-1,2-cyclopentanediol-3-methyl acetate-2-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (105 mg/fast diastereomer & 144 mg/slow diastereomer) as a colorless oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO fast δ (ppm) 7.43-6.86 (m, 13H, aromatic), 3.84 (m, 2H), 3.76 (m, 1H), 3.75 (m, 1H), 3.73 (s, 6H, 2 x OCH<sub>3</sub>), 3.57 (s, 3H, OCH<sub>3</sub>), 3.56-3.45 (m, 5H), 2.63-2.62 (m, 1H), 2.61-2.58 (m, 2H), 2.56-2.53 (m, 1H), 2.39 (d, 1H, J=9.5 Hz), 2.38 (d, 1H, J=9.5 Hz), 2.14 (m, 2H), 1.69 (m, 3H), 1.32 (m, 3H, CH<sub>2</sub>),

1.23 (m, 3H), 1.11 (d, 6H, J=6.8 Hz, 2 x CH<sub>3</sub>), 1.06 (d, 6H, J=6.8 Hz, 2 x CH<sub>3</sub>).

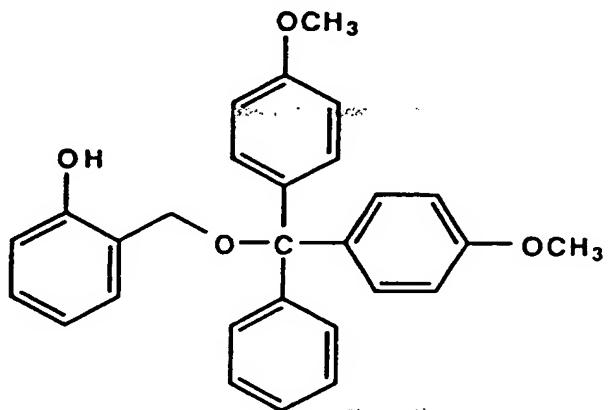
<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 148.8 fast.

<sup>1</sup>H NMR (500 MHz) DMSO slow δ (ppm) 7.43-6.84 (m, 13H, aromatic), 3.83-3.75 (m, 3H), 3.72 (s, 6H, 2 x OCH<sub>3</sub>), 3.65-3.59 (m, 2H), 3.57 (s, 3H, OCH<sub>3</sub>), 3.56-3.45 (m, 4H), 2.72 (m, 2H), 2.69 (m, 1H), 2.64-2.52 (m, 3H), 2.41-2.34 (m, 2H), 2.15 (m, 2H), 1.66 (m, 3H), 1.24 (m, 5H), 1.14-0.98 (m, 12H, 4 x CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 148.3. TLC R<sub>f</sub> 0.61 (9:1 trichloroethane/ethylacetate).

### Example 11

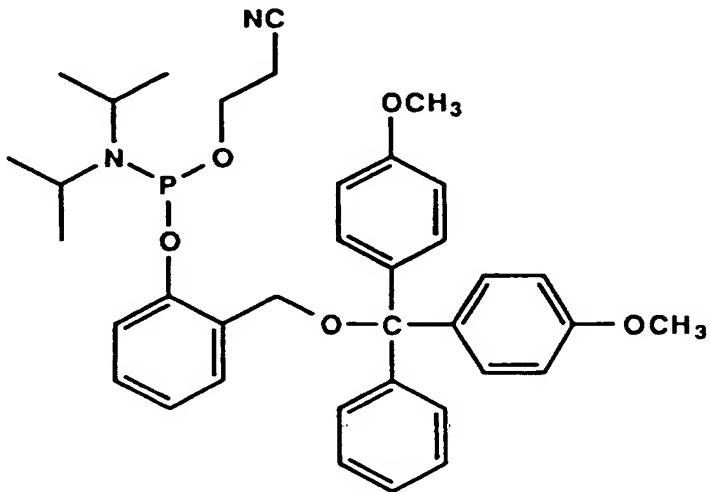
#### Synthesis of 2-(4,4'-dimethoxytrityl)-hydroxymethylphenol-1-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite

2-Hydroxybenzyl alcohol (5 g, 40.3 mmol) in dry pyridine (100 mL) was treated with DMT chloride (16.4 g, 48.3 mmol), triethylamine (5.5 g, 54.5 mmol), and dimethylaminopyridine (244 mg, 2 mmol) and stirred at room temperature for 19 hours. The mixture was poured into 5% aqueous sodium bicarbonate (500 mL), extracted with dichloromethane (4 x 100 mL). The combined organic layers were washed with saturated aqueous sodium chloride (100 mL) and dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness to give the crude product (20 g) which was purified by column chromatography on silica gel (300 g) eluting with a gradient of 1,1,1-trichloroethane/ethyl acetate (0 to 2%). The purities of the fractions were monitored by TLC and fractions containing the desired material were combined and evaporated to dryness to give 2-(4,4'-dimethoxytrityl)-hydroxymethylphenol (25.7 g) as a pale yellow oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 9.39 (s, 1H, phenolic OH), 7.59-6.72 (m, 17H, aromatic), 4.01 (s, 2H, benzylic CH<sub>2</sub>), 3.72 (s, 6H, 2 × OCH<sub>3</sub>), 3.68 (s, 3H, OCH<sub>3</sub>). TLC Rf 0.68 (1,1,1-trichloroethane).

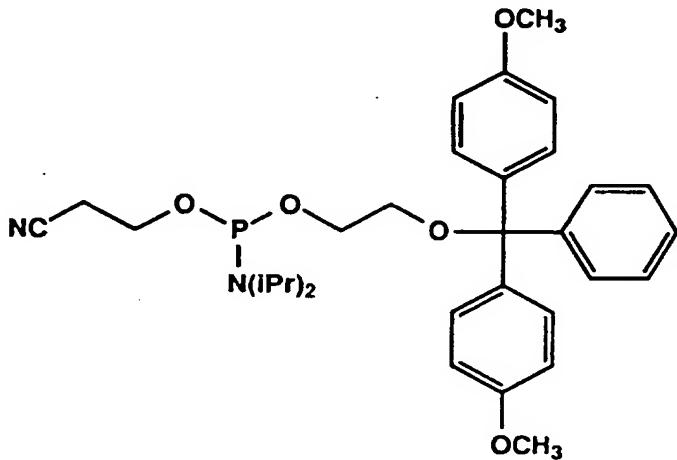
This DMT derivative (4 g, 10.5 mmol) was dissolved in dry dichloromethane (25 mL) under nitrogen and treated with diisopropylethylamine (6.7 g, 52 mmol), and 2-cyanoethoxy-(N,N-diisopropylamino)chlorophosphine (3.2 g, 13.5 mmol). After stirring for 1 hour, additional 2-cyanoethoxy-(N,N-diisopropylamino)-chlorophosphine (1.1 g, 4.2 mmol) was added and the mixture was stirred for 2 hours. Ethyl acetate (300 mL) and methanol (1 mL) were added and the mixture was washed with 10 % aqueous sodium carbonate (2 × 200 mL), aqueous sodium chloride (2 × 200 mL), add the organic layers were dried over anhydrous sodium sulfate. The solid was filtered off and the filtrate was evaporated to dryness and coevaporated with toluene to give the crude product. This was purified column chromatography on silica gel (200 g) eluting with dichloromethane. The purities of the fractions were monitored by TLC and fractions containing pure material were combined and evaporated to dryness to give 2-(4,4'-dimethoxytrityl)- hydroxymethylphenol-1-(N,N-diisopropylamino-2-cyanoethyl)-phosphoramidite (4.2 g, 6.8 mmol) as an oil having the following structure:



<sup>1</sup>H NMR (500 MHz) DMSO δ (ppm) 7.75-6.75 (m, 17H, aromatic), 4.10 (m, 2H), 4.01 (s, 1H, benzylic CH<sub>2</sub>), 3.72 (s, 6H, OCH<sub>3</sub>), 3.71-3.60 (m, 2H), 3.49 (m, 2H), 2.69 (m, 2H), 1.09 (d, 6H, J=6.7 Hz, 2 × CH<sub>3</sub>), 0.90 (d, 6H, J=6.7 Hz, 2 × CH<sub>3</sub>). <sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 145.2. TLC Rf 0.57, 0.49 (99:1 CH<sub>2</sub>Cl<sub>2</sub>/MeOH).

Example 12 $\Omega^1\text{-(2-Cyanoethyl-N,N-Diisopropyl-phosphoramidite)-O}^2\text{-(4,4'-Dimethoxytrityl)-2-Hydroxyethanol}$ 

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (0.68 mL, 0.04 mmol) was added dropwise to a stirred solution of  $\text{O}^2\text{-(4,4'-dimethoxytrityl)-2-hydroxyethanol}$  (1.3 g, 2.76 mmol) in anhydrous THF (15 mL) containing triethylamine (0.84 mL, 6.1 mmol). On addition, the mixture was stirred at RT for 30 min, and then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate-hexane mixture (3:7) and applied to a flash column. The product was eluted using 30% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the product as an almost colorless oil, (1.53 g, 82%) having the following structure:



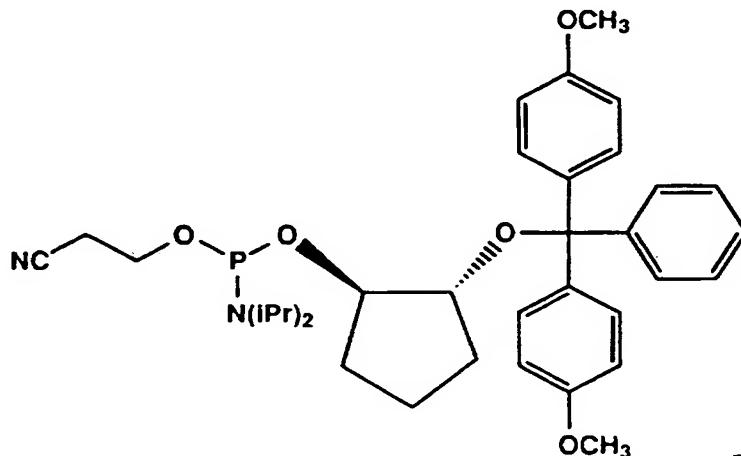
$^{31}\text{P}$  NMR  $\delta$  145.  $^1\text{H}$  NMR  $\delta$  7.50-6.80(13H, m aromtic), 3.70(6H, s,  $\text{OCH}_3$ ), 3.85-3.60(4H, m,  $\text{CH}_2\text{ODMT+CH}_2\text{OP+CH}_2\text{CH}_2\text{CN}$ ), 3.16(1H, m,  $\text{CH}_2\text{CN}$ ), 3.06(1H, m,  $\text{CH}_2\text{CN}$ ), 2.74(2H, m,  $\text{CH}(\text{CH}_3)_2$ ), 1.13(12H, m,  $\text{CH}_3$ ).  $R_f = 0.35$  ~30% Ethyl acetate in hexane.

Example 13

**O<sup>1</sup>-*(2-Cyanoethyl-N,N-Diisopropyl-phosphoramidite)-O<sup>2</sup>-*(4,4'-Dimethoxytrityl)-1,2-Dihydroxycyclopentane****

Dimethoxytrityl chloride (3.38 g, 10mmol) was added in portions to a stirred solution of the cyclopentanol (5 g, 49 mmol) in anhydrous pyridine (300mL) containing DMAP (20 mg). On addition, the mixture was left to stir at RT overnight after which the pyridine was removed under vacuum, and the residue adsorbed onto silica gel. This mixture was then applied to a flash silica gel column and the product eluted using 25% ethyl acetate in hexane. Homogenous fractions were combined and evaporated *in vacuo*, to give (+/-)-O<sup>2</sup>-4,4'-dimethoxytrityl-1,2- dihydroxycyclopentane as a yellow solid. R<sub>f</sub> = 0.29 (25% Ethyl acetate in hexane).

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (1.4 mL, 6.24 mmol) was added dropwise to a stirred solution of the above DMT derivative (2.29 g, 5.67 mmol) in anhydrous THF (20 mL) containing triethylamine (1.56 mL, 11.34 mmol). On addition, the mixture was stirred at RT for 60 min, and then filtered through a scintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was then eluted using 30% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the product as an almost colorless oil, (1.84 g, 57%) having the following structure:



<sup>31</sup>P NMR δ 147. <sup>1</sup>H NMR δ 7.50-6.80(13H, m, aromatic), 3.93(1H, m, CHOP), 3.82(1H, m, CHODMT), 3.72(6H, m, OCH<sub>3</sub>), 3.58(2H, m, CH<sub>2</sub>OP), 3.49(2H, m, CH<sub>2</sub>CN), 2.72(1H, m, NCH), 2.66(1H, m, NCH), 1.88(1H, m, alicyclic), 1.52(1H, m, alicyclic), 1.41(1H, m, alicyclic), 1.21(1H, m, alicyclic), 1.06(12H, m,

CH(CH<sub>3</sub>)<sub>2</sub>), 0.92(1H, m, alicyclic), 0.82(1H, m, alicyclic). R<sub>f</sub> = 0.47 (30% Ethyl acetate in hexane).

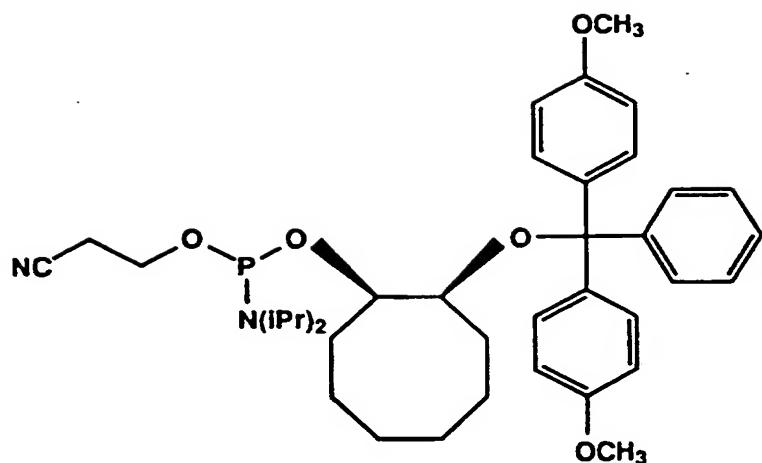
Example 14

Ω<sup>1</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-Ω<sup>2</sup>-(4,4'-Dimethoxytrityl)-1,2-Dihydroxycyclo-octane

Dimethoxytrityl chloride (3.76 g, 11.1 mmol) was added to a stirred solution of cis-1,2-dihydroxycyclooctane (8 g, 55.5 mmol) in anhydrous pyridine (300 mL) containing DMAP (20 mg). The reaction was left to stir at RT under nitrogen overnight. The reaction mixture was then evaporated under vacuum, and the residue adsorbed onto silica gel and applied to a silica gel flash column. The product was eluted using 30% ethyl acetate in hexane, and homogenous fractions combined and evaporated to give O<sup>2</sup>-(4,4'-dimethoxytrityl)-cis-1,2-dihydroxycyclooctane as an almost colorless oil, (3.30 g, 67%).

<sup>1</sup>H NMR δ 7.50-6.80 (13H, m, aromatic), 4.22 (1H, m, OH), 3.72 (6H, m, OCH<sub>3</sub>), 3.46 (2H, m, CHOH+CHODMT), 1.60-0.90 (12H, m, alicyclic). R<sub>f</sub> = 0.38 (30% Ethyl acetate in hexane).

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (0.54 mL, 2.4 mmol) was added dropwise to a stirred solution of the above DMT derivative (0.97 g, 2.18 mmol) in anhydrous THF (20 mL) containing triethylamine (0.63 mL, 4.59 mmol). On addition, the mixture was stirred at RT for 60 min, and then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was then eluted using 25% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as an almost colorless oil, (1.07 g, 80%) with the following structure:



$^{31}\text{P}$  NMR  $\delta$  145.  $^1\text{H}$  NMR  $\delta$  7.50-6.80 (13H, m, aromatic), 3.72 (6H, s, OCH<sub>3</sub>), 3.63 (2H, m, CH<sub>2</sub>OP), 3.48 (2H, m, CH<sub>2</sub>CN), 2.68 (2H, m, NCH), 1.70-1.20 (12H, m, alicyclic), 1.10 (6H, m, CH(CH<sub>3</sub>)<sub>2</sub>), 1.00 (6H, m, CH(CH<sub>3</sub>)<sub>2</sub>). R<sub>f</sub> = 0.27 (25% Ethyl acetate in hexane).

#### Example 15

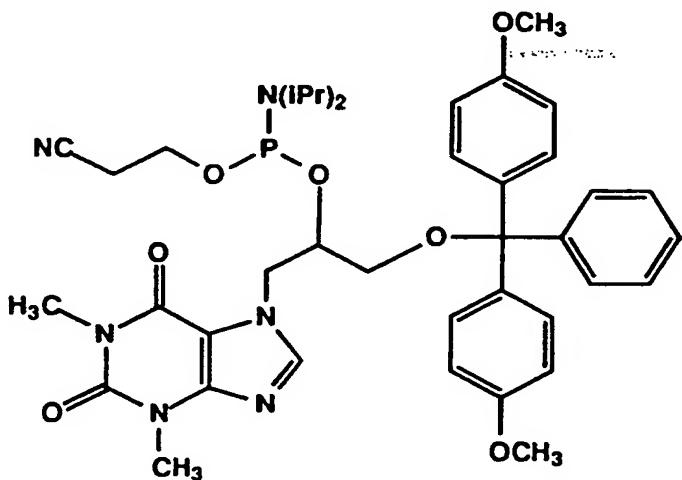
#### Q<sup>2</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-Q<sup>3</sup>-(4,4'-Dimethoxytrityl)-7-(2,3-Dihydroxypropyl) theophylline

Dimethoxytrityl chloride (22.7 g, 67 mmol) was added in portions to a stirred solution of 7-(2,3-Dihydroxypropyl)theophylline (10 g, 61 mmol) in anhydrous pyridine (200 mL) containing DMAP (20 mg). The reaction was left to stir at RT under nitrogen overnight. The reaction mixture was then evaporated under vacuum, and the residue adsorbed onto silica gel and applied to a silica gel flash column. The product was eluted using 3% methanol in dichloromethane, and homogenous fractions combined and evaporated to give 7-(O<sup>3</sup>-(4,4'-dimethoxytrityl)-2,3-dihydroxypropyl)theophylline as an almost colorless oil (14.4 g, 51%).

$^1\text{H}$  NMR  $\delta$  7.91(1H, s, H8), 7.50-6.80(13H, m, aromatic), 5.28(1H, d, CHOH), 4.47(1H, d of d, NCH<sub>2</sub>), 4.16(1H, d of d, NCH<sub>2</sub>), 4.04(1H, m, CHOH), 3.72(6H, s, OCH<sub>3</sub>), 3.40(3H, s, NCH<sub>3</sub>), 3.22(3H, s, NCH<sub>3</sub>), 3.00(1H, m, CH<sub>2</sub>ODMT), 2.87(1H, m, CH<sub>2</sub>ODMT). R<sub>f</sub> = 0.29 (3% Methanol in dichloromethane).

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (4.6 mL, 20 mmol) was added dropwise to a stirred solution of the above DMT derivative (5.0 g, 9 mmol) in anhydrous THF (40 mL) containing triethylamine (2.7 mL, 19 mmol). On addition, the mixture was stirred at RT for 60 min, and then filtered through a sintered glass funnel

under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was then eluted using 35% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as a powder, (3.9 g, 57%) with the following structure:



<sup>31</sup>P NMR δ 149. <sup>1</sup>H NMR δ 7.99 (1H, s, H8), 7.50-6.80 (13H, m, aromatic), 4.50 (2H, m, N7CH<sub>2</sub>), 4.38 (1H, m, CHOP), 3.72 (6H, s, OCH<sub>3</sub>), 3.51 (2H, m, OCH<sub>2</sub>), 3.42 (2H, m, CH<sub>2</sub>CN), 3.40 (3H, s, NCH<sub>3</sub>), 3.22 (3H, s, NCH<sub>3</sub>), 3.18 (1H, m, CH<sub>2</sub>ODMT), 3.10 (1H, m, CH<sub>2</sub>ODMT), 2.55 (2H, m, NCH), 1.05 (6H, d, CHCH<sub>3</sub>), 0.97 (6H, d, CHCH<sub>3</sub>). Rf = 0.34, 0.16 (50% Ethyl acetate in hexane).

#### Example 16

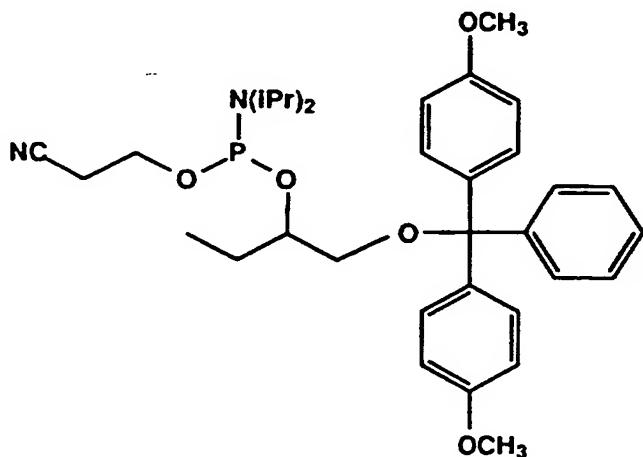
#### (+/-)-O<sup>2</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-O<sup>1</sup>-(4,4'-Dimethoxytrityl)-1,2-Dihydroxybutane

4,4'-Dimethoxytrityl chloride (10.33 g, 30.5 mmol) was added in portions to a stirred solution of 1,2-butanediol (2.5 g, 27.7 mmol) in anhydrous pyridine (200 mL) containing DMAP (1.7 g). The mixture was stirred at RT under nitrogen overnight. The solution was then evaporated under vacuum, and the residue adsorbed onto silica gel and applied to a silica gel flash column which was eluted using 10% ethyl acetate in hexane. Homogenous fractions were combined and evaporated to give (+/-)-O<sup>1</sup>-(4,4'-dimethoxytrityl)-1,2-dihydroxybutane as an almost colorless oil, (8.15 g, 75%).

<sup>1</sup>H NMR δ 7.50-6.80 (13H, m, aromatic), 4.59 (1H, d, OH), 3.73 (6H, s, OCH<sub>3</sub>), 3.53 (1H, d of d, CHOH), 2.91 (1H, t, CH<sub>2</sub>ODMT), 2.78 (1H, t,

$\text{CH}_2\text{ODMT}$ ), 1.55 (1H, m,  $\text{CH}_2\text{CH}_3$ ), 1.29 (1H, m,  $\text{CH}_2\text{CH}_3$ ), 0.8 (3H, t,  $\text{CH}_3$ ).  $R_f$  = 0.2 (10% Ethyl acetate in hexane).

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (0.94 mL, 4.2 mmol) was added dropwise to a stirred solution of the above DMT derivative (1.5 g, 3.8 mmol) in anhydrous THF (20 mL) containing triethylamine (1.12 mL., 8.4 mmol). On addition, the mixture was stirred at RT for 7 h., and was then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate hexane mixture (3:7) and applied to a flash column. The product was then eluted using 20% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the product as a viscous oil, (1.0 g, 44%) with the following structure:



$^{31}\text{P}$  NMR  $\delta$  148.5.  $^1\text{H}$  NMR  $\delta$  7.50-6.80(13H, m, aromatic), 3.82(1H, m,  $\text{OCH}_3$ ), 3.71(6H, s,  $\text{OCH}_3$ ), 3.53(2H, m,  $\text{CH}_2\text{CN}$ ), 2.97(1H, m,  $\text{CH}_2\text{ODMT}$ ), 2.80(1H, m,  $\text{CH}_2\text{ODMT}$ ), 2.60(1H, m, NCH), 1.67(1H, m,  $\text{CH}_3\text{CH}_2$ ), 1.51(1H, m,  $\text{CH}_3\text{CH}_2$ ), 1.12(12H, m,  $\text{CH}(\text{CH}_3)_2$ ), 0.78(3H, m,  $\text{CH}_2\text{CH}_3$ ).  $R_f$  = 0.16 20% Ethyl acetate in hexane.

#### Example 17

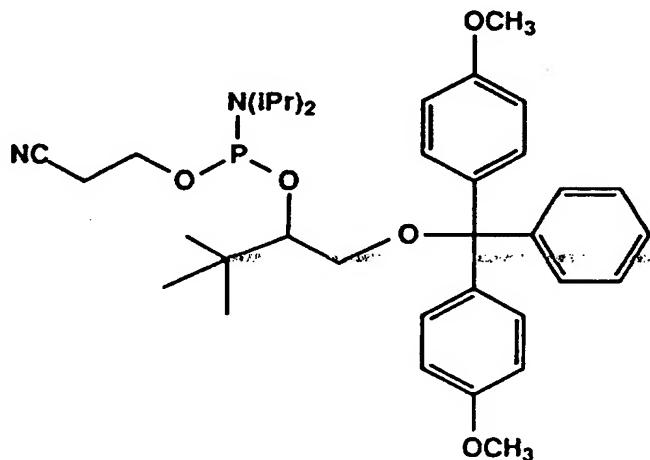
#### (+/-)-O<sup>2</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-O<sup>1</sup>-(4,4'-Dimethoxytrityl)-3,3-Dimethyl-1,2-Dihydroxybutane

4,4'-Dimethoxytrityl chloride (9.4 g, 27.7 mmol) was added in portions to a stirred solution of freshly distilled 3,3 dimethyl-1,2-dihydroxybutanediol (3 g, 25.4 mmol.) in anhydrous pyridine (200 mL) containing DMAP (0.31 g). The mixture was

stirred at RT under nitrogen overnight. The solution was then evaporated under vacuum, and the residue adsorbed onto silica gel and applied to a silica gel short path column. The product was eluted using 2L of 10% ethyl acetate followed by 1L of 20% ethyl acetate in hexane, and finally 1L of 30% ethyl acetate in hexane. Homogenous fractions were combined and evaporated to give (+/-)-O<sup>1</sup>- (4,4'-dimethoxytrityl)-3,3-dimethyl-1,2-dihydro xybutane as an almost colorless oil, (6.03 g, 56.5%).

<sup>1</sup>H NMR δ 7.50-6.70(13H, m aromatic), 4.70(1H, d, OH), 3.72(6H, s, OCH<sub>3</sub>), 3.32(1H, t, CHOH), 2.93(2H, m, CH<sub>2</sub>), 0.72(9H, s, CH<sub>3</sub>). Rf = 0.37 (20% Ethyl acetate in hexane).

2-Cyanoethyl-N,N-diisopropylchlorophosphoramidite (1.81 mL, 8.1 mmol) was added dropwise to a stirred solution of the above DMT derivative (2.27 g, 5.4 mmol) in anhydrous THF (50 mL) containing triethylamine (1.6 mL, 11.3 mmol). On addition, the mixture was stirred at RT overnight, and was then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was then evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was eluted using 30% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as a viscous oil, (1.89 g, 56%) having the following structure:



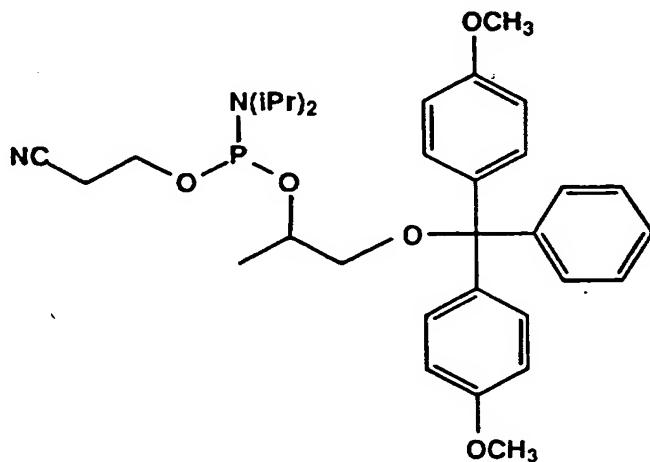
<sup>31</sup>P NMR δ 149. <sup>1</sup>H NMR δ 7.60-6.80 (13H, m, aromatic), 3.82 (1H, m, CHOP), 3.65 (2H, m, OCH<sub>2</sub>), 3.63 (2H, m, CH<sub>2</sub>CN), 3.19 (1H, d of d, CH<sub>2</sub>ODMT), 3.08 (1H, d of d, CH<sub>2</sub>ODMT), 2.79 (1H, m, NCH), 2.67 (1H, m, NCH), 1.15 (12H, m, CH(CH<sub>3</sub>)<sub>2</sub>), 0.78 (9H, m, (CH<sub>3</sub>)<sub>3</sub>). Rf = 0.13 (20% Ethyl acetate in hexane).

Example 18(+/-)-O<sup>2</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-O<sup>1</sup>-(4,4'-Dimethoxytrityl)-1,2-Dihydroxypropane

4,4'-Dimethoxytrityl chloride (7.42 g, 21.9 mmol) was added in portions to a stirred solution of 1,2-propanediol (1.52 g, 20.0 mmol) in anhydrous pyridine (200 mL) containing DMAP (0.28 g). The mixture was stirred at RT under nitrogen overnight. The solution was then evaporated under vacuum, and the residue adsorbed onto silica gel and applied to a silica gel flash column. The product was eluted using 20% ethyl acetate in hexane, and homogenous fractions were combined and evaporated to give (+/-)-O<sup>1</sup>-(4,4'-dimethoxytrityl)-1,2-dihydroxypropane as an almost colorless oil(3.1 g, 41%).

<sup>1</sup>H NMR δ s 7.50-6.80 (13H, m, aromatic), 4.6 (1H, d, CHO<sub>H</sub>), 3.78 (1H, m, CHO<sub>H</sub>), 3.66 (6H, s, OCH<sub>3</sub>), 2.91 (1H, t, CH<sub>2</sub>), 2.68 (1H, t, CH<sub>2</sub>). R<sub>f</sub> = 0.25 (20% Ethyl acetate in hexane).

2-Cyanoethyl-N,N-Diisopropylchlorophosphoramidite (1.36 mL, 6.1 mmol) was added dropwise to a stirred solution of the above DMT derivative (2.1 g, 5.6 mmol) in anhydrous THF (50 mL) containing triethylamine (1.63 mL., 11.7 mmol). On addition, the mixture was stirred at RT overnight., and was then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was then eluted using 20% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as a viscous oil, (1.0 g, 31%) having the following structure:



<sup>31</sup>P NMR δ 147. <sup>1</sup>H NMR δ 7.50-6.70 (13H, m, aromatic), 4.03 (1H, m, CH<sub>OP</sub>), 3.72 (6H, s, OCH<sub>3</sub>), 3.65 (2H, m, OCH<sub>2</sub>), 3.58 (2H, m, CH<sub>2</sub>CN), 3.05 (1H, m, CH<sub>2</sub>ODMT), 2.85 (1H, m, CH<sub>2</sub>ODMT), 2.77 (1H, m, NCH), 2.64 (1H, m, NCH), 1.15 (12H, m, NCH(CH<sub>3</sub>)<sub>2</sub>). R<sub>f</sub> = 0.28, 0.33 (20% Ethyl acetate in hexane).

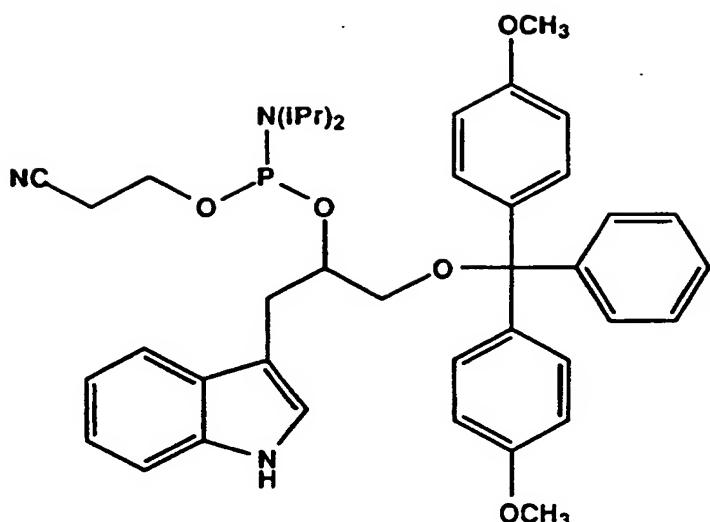
### Example 19

#### (+/-)-O<sup>2</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-O<sup>1</sup>-(4,4'-Dimethoxytrityl)-3-(3-Indolyl)-1,2-Dihydroxy propane

4,4'-Dimethoxytrityl chloride (1.7 g, 5.0 mmol) was added in portions to a stirred solution of (+/-)-3-(3-indolyl)-1,2 dihydroxypropane (0.87 g, 4.6 mmol) in anhydrous pyridine (150 mL) containing DMAP (0.28 g.). On addition, the solution was left to stir at RT overnight under nitrogen, after which the mixture was evaporated under vacuum. The residue was then adsorbed onto silica gel, and this mixture applied to a silica gel short path column and the product eluted using 1% methanol in dichloromethane. Homogenous fractions were combined, and evaporated *in vacuo* to give the (+/-)-O<sup>1</sup>-(4,4'-dimethoxytrityl)-3-(3-indolyl)-1,2 dihydroxypropane as an oil(0.5 g, 22%).

<sup>1</sup>H NMR δ 7.60-6.75 (18H, m, aromatic), 4.64 (1H, d, ex., OH), 3.92 (1H, m, CHOH), 3.72 (6H, s, OCH<sub>3</sub>), 2.94 (3H, m, CH<sub>2</sub>-ind.+CH<sub>2</sub>-OH), 2.72 (1H, m, CH<sub>2</sub>OH). R<sub>f</sub> = 0.25 (1% Methanol in dichloromethane).

2-Cyanoethyl-N,N-diisopropylchlorophosphoramidite (0.18 mL, 0.81 mmol) was added dropwise to a stirred solution of the above DMT derivative (0.4 g, 0.74 mmol) in anhydrous THF (25 mL) containing triethylamine (0.22 mL, 1.55 mmol). On addition, the mixture was stirred at RT overnight., and then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was eluted using 50% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the product as a viscous oil, (0.224g, 41%) having the following structure:

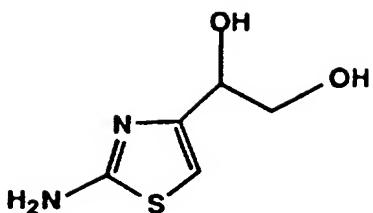


<sup>31</sup>P NMR δ 148. <sup>1</sup>H NMR δ 7.50-6.70 (18H, m, aromatic), 4.24 (1H, m, CHOH), 3.72 (6H, m, OCH<sub>3</sub>), 3.70-3.42 (4H, m, CH<sub>2</sub>CN+CH<sub>2</sub>OP), 3.18-2.90 (CH<sub>2</sub>ODMT+CH<sub>2</sub>-ind), 1.28-0.93 (12H, m, CH(CH<sub>3</sub>)<sub>2</sub>). Rf = 0.47 (20% Ethyl acetate in hexane).

#### Example 20

(+/-)-O<sup>1</sup>-(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)-O<sup>2</sup>-(4',4'-Dimethoxytrityl)-N-Acetyl-2-Amino-4-(1,2-Dihydroxyethyl)-1,3-Thiazole

A solution of ethyl 2-amino-4-thiazolglyoxylate (10g, 50mmol) in anhydrous THF (150 mL) was added dropwise to a suspension of lithium aluminum hydride (1.9 g) in anhydrous THF (200 mL), and the mixture stirred under nitrogen at RT for 1 h. Excess ethyl acetate was then added to destroy residual reductant, after which an excess of Glauber's Salt was added and the resultant suspension stirred at RT for 40min. The slurry was then filtered, and the filtrate evaporated *in vacuo*, and then dissolved in methanol and adsorbed onto silica gel. This mixture was then applied to a flash silica column, and the product eluted using 10% methanol in dichloromethane. Homogenous fractions were combined and evaporated under vacuum to give (+/-)-2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole product as a pale orange colored solid, (2.6 g, 32.5%) with the following structure:



<sup>1</sup>H NMR δ 6.74 (2H, s, NH<sub>2</sub>), 6.28 (1H, s, aromatic), 4.92 (1H, broad, OH), 4.48 (1H, broad, OH), 4.34 (1H, m, CHOH), 3.60 (1H, d of d, CH<sub>2</sub>OH), 3.38 (1H, m, CH<sub>2</sub>OH). Rf = 0.43 (30 % Methanol in dichloromethane).

Acetic anhydride (4.5 mL, 47.4 mmol) was added to a solution of the above aminothiazole (2.3 g, 14.4 mmol) in anhydrous pyridine (30 mL) and the solution stirred under nitrogen at RT overnight. The mixture was then evaporated under vacuum, and the residue dissolved in ethly acetate. This solution was washed successively with 2M hydrochloric acid, saturated sodium bicarbonate solution, and then brine. The mixture was then dried over anhydrous sodium sulfate, filtered and the filtrate evaporated under reduced pressure. The residue was triturated with ether, after which the product separated as a tan colored solid which was collected by filtration, giving (+/-)-N<sup>2</sup>,O,O-triacetyl-2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole as a powder, (1.08 g, 37%).

<sup>1</sup>H NMR δ 7.14 (1H, s, aromatic), 5.96 (1H, m, CHOAc), 4.42 (1H, m, CH<sub>2</sub>OAc), 4.33 (1H, m, CH<sub>2</sub>OAc), 2.12 (3H, s, CH<sub>3</sub>CO), 2.07 (3H, s, CH<sub>3</sub>CO), 1.98 (3H, s, CH<sub>3</sub>CONH). Rf = 0.48 (10% Methanol in dichloromethane).

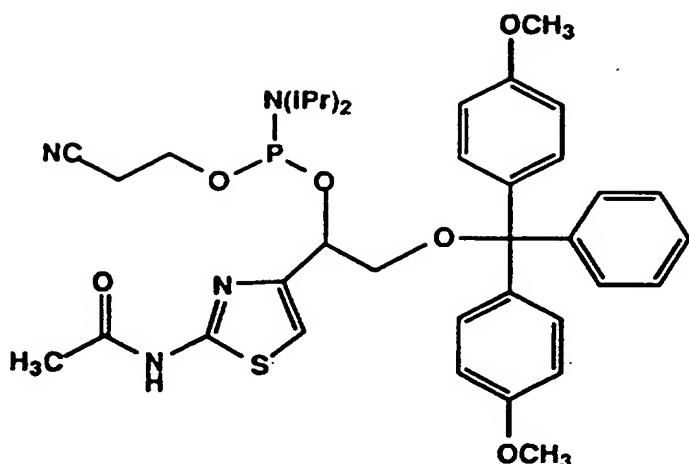
The above triacetylthiazole derivative (1.0 g, 3.7 mmol.) was suspended in methanol and 0.1 M sodium hydroxide solution (20 mL) was added in a single portion. The reaction mixture was stirred at RT and the reaction carefully monitored by TLC (10% methanol in dichloromethane). The reaction was stopped after 40' min by neutralizing the mixture using 2M hydrochloric acid in an ice bath. The resultant mixture was evaporated to dryness under vacuum, and the residue dissolved in a mixture of dichloromethane and methanol and then adsorbed onto silica gel. This material was applied to a flash silica column, and the product eluted using 15% methanol in dichloromethane. Homogenous fractions were combined and evaporated *in vacuo* to give (+/-)-N-acetyl-2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole as a tan colored powder (0.475 g, 69%).

<sup>1</sup>H NMR δ 6.88 (1H, s, aromatic), 5.12 (1H, d, OH), 4.53 (1H, m, CHOH), 3.66 (1H, m, CH<sub>2</sub>), 3.47 (1H, m, CH<sub>2</sub>), 2.11 (3H, s, CH<sub>3</sub>CO). Rf = 0.22 (10% Methanol in dichloromethane).

The N-acetyl-derivatized thiazole derivative from the above reaction (0.915 g, 2.7 mmol.) was added in a single portion to a stirred solution of N-acetyl- 2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole in anhydrous pyridine (20 mL.) containing DMAP (20 mg). The resultant mixture was then left to stir at RT under nitrogen for 3 h, and was then evaporated under vacuum to give an orange colored oil. This material was dissolved in dichloromethane and adsorbed onto silica gel and then applied to a silica flash column. The product was eluted using 3% methanol in dichloromethane, and homogenous fractions combined and evaporated *in vacuo* to give an off-white foam of (+/-)-O<sup>2</sup>-(4,4'-dimethoxytrityl)-N-acetyl-2-amino-4-(1,2-dihydroxyethyl)-1,3-thiazole (800 mg, 66%).

<sup>1</sup>H NMR δ 7.44-6.80 (14H, m, aromatic), 5.42 (1H, d, OH), 4.79 (1H, m, CHOH), 3.73 (6H, s, OCH<sub>3</sub>), 3.18 (1H, m, CH<sub>2</sub>ODMT), 3.12 (1H, m, CH<sub>2</sub>ODMT).  
Rf = 0.34 (5% Methanol in dichloromethane).

2-Cyanoethyl-N,N-diisopropylchlorophosphoramidite (1.0 mL, 4.48 mmol) was added dropwise to a stirred solution of the above DMT derivative (1.0 g, 2.04 mmol) in anhydrous THF (45 mL) containing triethylamine (1.2 mL, 8.56 mmol). On addition, the mixture was stirred at RT for 4 h, and then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate hexane mixture (3:7) and applied to a flash column. The product was then eluted using 50% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as a viscous oil, (0.224g, 41%) with the following structure:



$^{31}\text{P}$  NMR  $\delta$  149.  $^1\text{H}$  NMR  $\delta$  12.09 (1H, s, NH), 7.50-6.7 (14H, m, aromatic), 4.96 (1H, m,  $\text{CHOP}$ ), 3.72 (6H, s,  $\text{OCH}_3$ ), 3.62 (4H, m,  $\text{CH}_2\text{OP} + \text{CH}_2\text{CN}$ ), 3.30 (1H, m,  $\text{CH}_2\text{ODMT}$ ), 3.09 (1H, m,  $\text{CH}_2\text{ODMT}$ ), 2.06 (3H, s,  $\text{CH}_3\text{CO}$ ), 1.25-1.04 (12H, m,  $\text{CH}(\text{CH}_3)_2$ ). Rf = 0.52, 0.58 (50% Ethyl acetate in hexane).

### Example 21

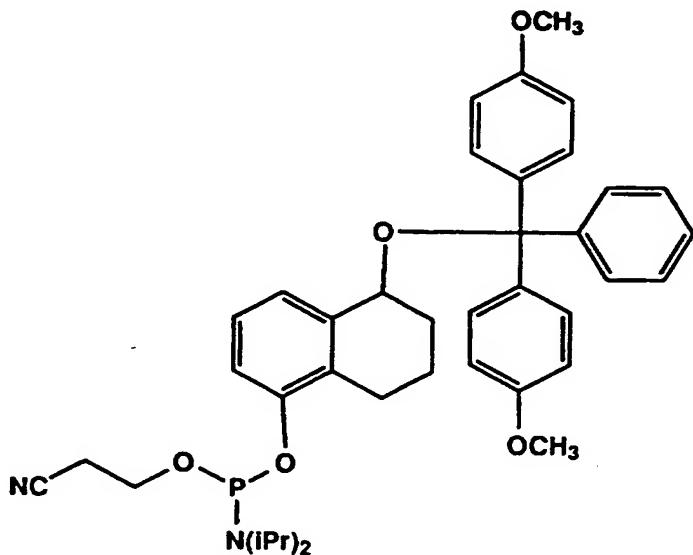
#### $\text{O}^5$ -(2-Cyanoethyl-N,N-Diisopropylphosphoramidite)- $\text{O}^1$ -(4,4'-Di methoxytrityl)-1,2,3,4-Tetrahydro-1,5-Dihydroxynaphthalene

Dimethoxytrityl chloride (11.35 g, 33.5 mmol) was added to a stirred solution of 1,5-dihydroxy-1,2,3,4-tetrahydronaphthalene (5 g, 30.5 mmol) in pyridine (100 mL) containing DMAP (200 mg), and the resultant mixture was stirred for 2 h at room temperature under nitrogen. The reaction mixture was then poured into water and the product extracted with ethyl acetate. The extracts were washed with brine, dried over anhydrous sodium sulfate, filtered, and the filtrate evaporated *in vacuo*. The residue was then adsorbed onto silica gel, and the mixture applied to a silica gel short path column and the product eluted using 30% ethyl acetate in hexane. Homogenous fractions were combined and evaporated *in vacuo* to give  $\text{O}^1$ -(4,4'-dimethoxytrityl)-1,2,3,4-tetrahydro-1,5-dihydroxy naphthalene as an off-white foam, (5.52 g, 38%).

$^1\text{H}$  NMR  $\delta$  9.01 (1H, s, OH), 7.50-6.50 (16H, m, aromatic), 4.21 (1H, m,  $\text{CHOH}$ ), 3.72 (6H, s,  $\text{OCH}_3$ ), 2.60 (1H, m,  $\text{CH}_2$ -aromatic), 2.34 (1H, m,  $\text{CH}_2$ -aromatic), 1.88 (1H, m, alicyclic), 1.37 (1H, m, alicyclic), 1.28 (1H, m, alicyclic), 1.23 (1H, m, alicyclic). Rf = 0.17 (30% Ethyl acetate in Hexane).

2-Cyanoethyl-N,N-diisopropylchlorophosphoramidite (0.56 mL, 2.5 mmol) was added dropwise to a stirred solution of the above DMT derivative (1.0 g, 2.3 mmol) in anhydrous THF (10 mL) containing diisopropylethylamine (0.88 mL, 5.1 mmol). On

addition, the mixture was stirred at RT for 1 h, and was then filtered through a sintered glass funnel under nitrogen. The filtrate was evaporated under vacuum, and the residue dissolved in anhydrous benzene and then filtered. The filtrate was evaporated *in vacuo*, and the residue dissolved in an ethyl acetate/hexane mixture (3:7) and applied to a flash column. The product was then eluted using 30% ethyl acetate in hexane containing 1% triethylamine, and homogenous fractions were combined and evaporated to give the title compound as a viscous oil (0.92 g, 63%), having the following structure:



$^{31}\text{P}$  NMR  $\delta$  145.6.  $^1\text{H}$  NMR  $\delta$  7.50-6.80 (16H, m, aromatic), 4.23 (1H, m, CHOP), 3.93-3.60 (10H, m, OCH<sub>3</sub>+CH<sub>2</sub>OP+CH<sub>2</sub>CN), 2.80 (2H, m, NCH), 2.70 (1H, m, benzylic CH), 2.42 (1H, m, benzylic CH), 1.88 (1H, m, alicyclic), 1.42 (1H, m, alicyclic), 1.31 (1H, m, alicyclic), 1.09 (13H, m, alicyclic + CH(CH<sub>3</sub>)<sub>2</sub>). Rf = 0.42 (30% Ethyl acetate in hexane).

### Example 22

#### Synthesis of 5'-O-(4,4'-dimethoxytrityl)-diol-triethylammonium-H-phosphonates

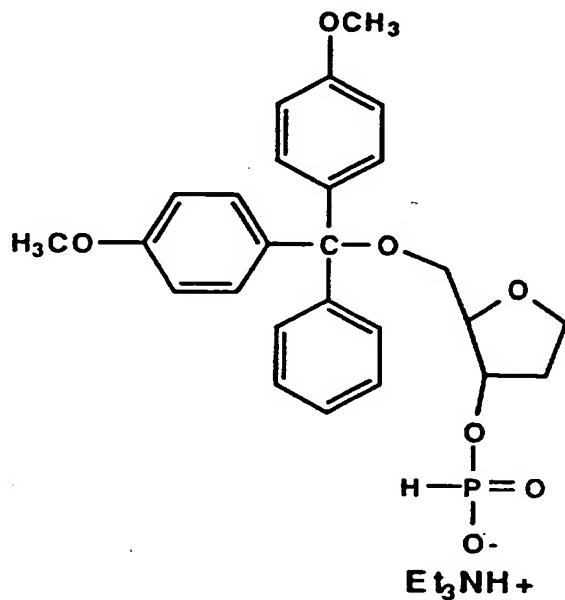
The DMT derivatives from examples 1 to 21 are dissolved in dry dichloromethane and added a flask over 10 minutes containing imidazole (15 equivalents), phosphorous trichloride (4.3 equivalents), triethylamine (29 equivalents), and dry dichloromethane cooled in an ice bath. The reaction mixture is stirred for 30 minutes at 0 °C, the ice bath is removed and the mixture is stirred for an additional 30 minutes. Water is added and this mixture was stirred for 10 minutes. The layers are separated and the aqueous layer is extracted with chloroform. The combined organic layers are evaporated and then co-evaporated twice with toluene. The crude material is purified by column

chromatography on silica gel eluting with a gradient of dichloromethane/methanol (0 to 30%). The purities of the fractions are monitored by thin layer chromatography (TLC) and fractions containing pure material are combined and evaporated to dryness. The resulting material is dissolved in dichloromethane and washed with 0.1 M triethylammonium bicarbonate. The aqueous layer is backwashed once with dichloromethane and the combined organic layers are evaporated to dryness to give the H-phosphonate monomer.

### Example 23

#### Synthesis of 5'-O-(4,4'-dimethoxytrityl)-1,2-dideoxy-D-ribose-3'-H-phosphonate

5'-(4,4'-dimethoxytrityl)-1,2-dideoxy-D-ribose (5.6 g, 13.4 mmol) was treated according to the protocol used in example 22 to give 5'-O-(4,4'-dimethoxytrityl)-1,2-dideoxy-D-ribose-3'-H-phosphonate (6.3 g) as a colorless foam having the following structure:

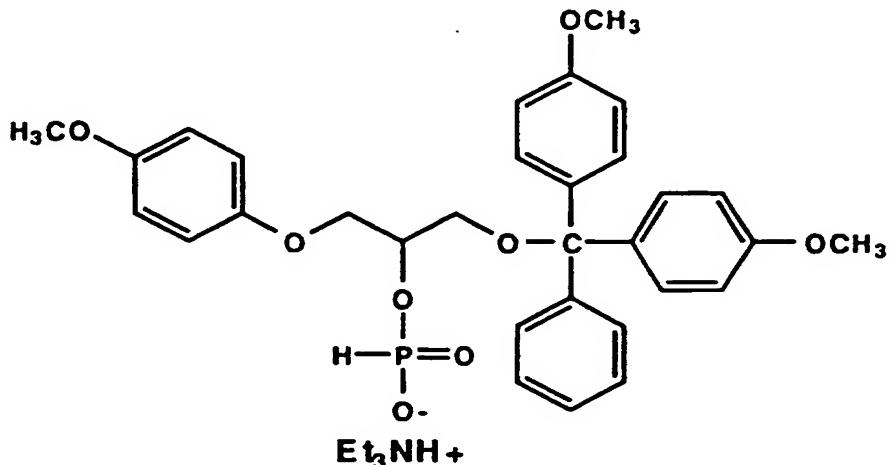


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 0.61 (doublet of doublets,  $J_{\text{P}-\text{H}} = 578 \text{ Hz}$ ,  $J_{\text{P}-\text{C}-\text{H}} = 9.1 \text{ Hz}$ ). TLC  $R_f = 0.37$  (8:2 dichloromethane/methanol/0.1% triethylamine).

### Example 24

#### Synthesis of 1-(4,4'-dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol-2-H-phosphonate

1-(4,4'-Dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol (3 g, 6 mmol) was treated according to the protocol used in example 22 to give 1-(4,4'-dimethoxytrityl)-3-(4-methoxyphenoxy)-1,2-propanediol-2-H-phosphonate (0.65 g) as a colorless oil having the following structure:

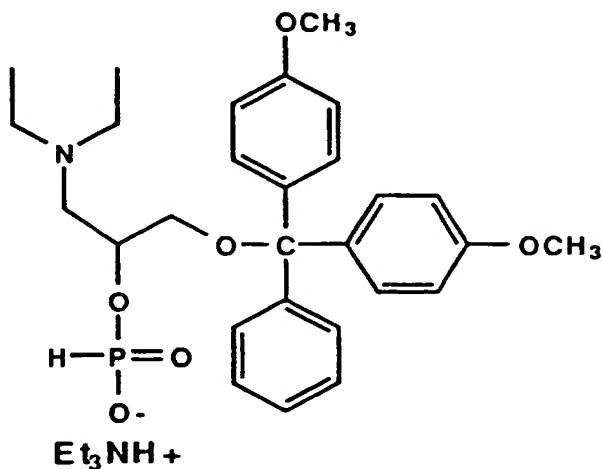


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.5 (doublet of doublets,  $J_{\text{P},\text{H}} = 586$  Hz,  $J_{\text{P},\text{O}-\text{C}-\text{H}} = 10.7$  Hz). TLC  $R_f = 0.26$  (9/1 dichloromethane/methanol).

### Example 25

#### Synthesis of 1-(4,4'-dimethoxytrityl)-3-(diethylamino)-1,2-propanediol-2-H-phosphonate

1-(4,4'-Dimethoxytrityl)-3-(diethylamino)-1,2-propanediol (2.7 g, 6 mmol) was treated according to the protocol used in example 22 to give 1-(4,4'-dimethoxytrityl)-3-(diethylamino)-1,2-propanediol-2-H-phosphonate (4.1 g) as an oil having the following structure:

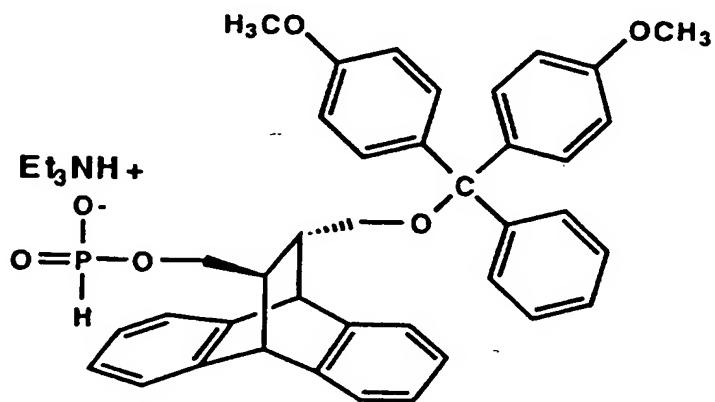


<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 5.2 (dd, J<sub>P-H</sub> = 601 Hz, J<sub>P-O-C-H</sub> = 10.7 Hz). TLC R<sub>f</sub> 0.26 (9:1 dichloromethane/methanol).

### Example 26

#### Synthesis of 4,4'-dimethoxytrityl-trans-9,10-ethanoanthracene-11,12-dimethanol-H-phosphonate

4,4'-Dimethoxytrityl-*trans*-9,10-ethanoanthracene-11,12-dimethanol (1.64 g, 2.9 mmol) was treated according to the protocol used in example 22 to give 4,4'-dimethoxytrityl-*trans*-9,10-ethanoanthracene-11,12-dimethanol-H-phosphonate (1.8 g) as a white foam having the following structure:

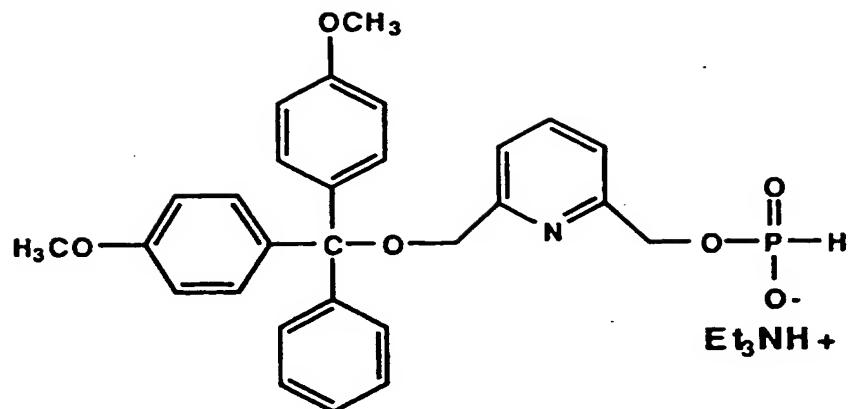


<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 1.6 (J<sub>P-H</sub> = 613 Hz). TLC R<sub>f</sub> = 0.09 (dichloromethane/0.5% triethylamine).

### Example 27

#### Synthesis of 2-(4,4'-dimethoxytrityl)-2,6-bis-hydroxymethylpyridine-6-H-phosphonate

2-(4,4'-Dimethoxytrityl)-2,6-bis-hydroxymethylpyridine (7.0 g, 15.8 mmol) was treated according to the protocol used in example 22 to give 2-(4,4'-dimethoxytrityl)-2,6-bis-hydroxymethylpyridine-6-H-phosphonate (8 g) as a yellow oil having the following structure:

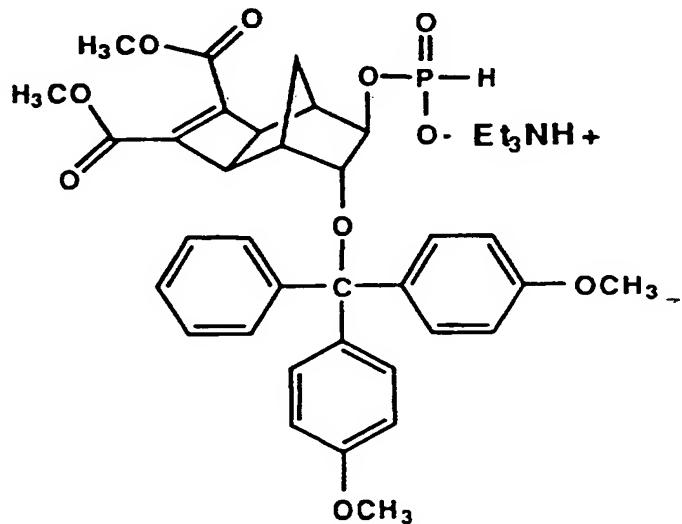


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.65 (doublet of triplets,  $\text{J}_{\text{P}-\text{H}} = 584$  Hz,  $\text{J}_{\text{P}-\text{O}-\text{CH}_2} = 9.1$  Hz). TLC  $R_f$  0.08 (9:1 dichloromethane/methanol/0.1% triethylamine).

### Example 28

**Synthesis of 7-(4,4'-dimethoxytrityl)-trans-7,8-dihydroxy-dimethyl-exo-tricyclononene[4.2.1.0^2,5]nona-3-ene-3,4-dicarboxylate-8-H-phosphonate**

7-(4,4'-Dimethoxytrityl)-trans-7,8-dihydroxy-dimethyl-exo-tricyclononene [4.2.1.0<sup>2,5</sup>]nona-3-ene-3,4-dicarboxylate (3.5 g, 6.1 mmol) was treated according to the protocol used in example 22 to give 7-(4,4'-dimethoxytrityl)-trans-7,8-dihydroxy-dimethyl-exo-tricyclononene[4.2.1.0<sup>2,5</sup>]nona-3-ene-3,4-dicarboxylate-8-H-phosphonate (4.8 g) as a white foam having the following structure:

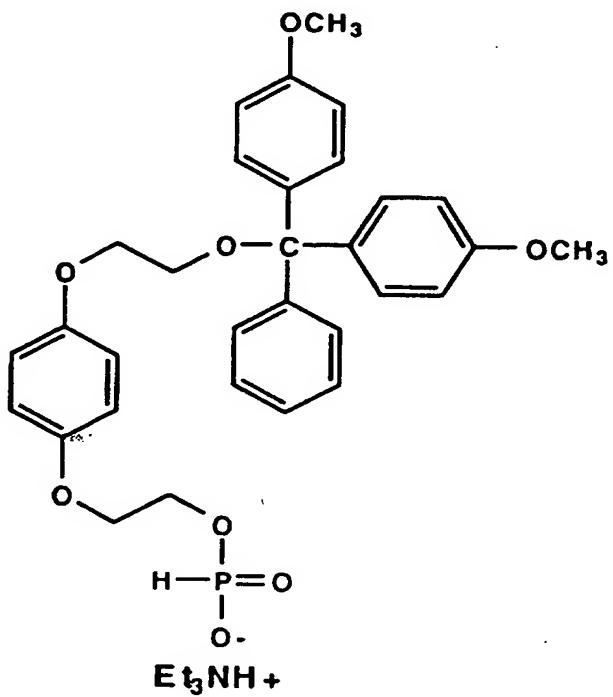


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 0.66 ( $J_{\text{P}-\text{H}} = 613$  Hz). TLC Rf 0.12 (9:1 dichloromethane/methanol/0.1% triethylamine).

### Example 29

#### Synthesis of (4,4'-dimethoxytrityl)-hydroquinone-bis-(hydroxyethyl)-ether-H-phosphonate

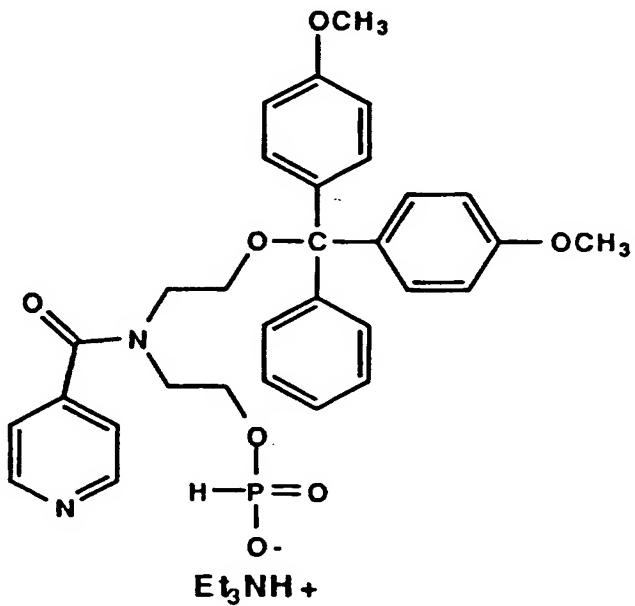
(4,4'-Dimethoxytrityl)-hydroquinone-bis-(hydroxyethyl)-ether (2.2 g, 4.4 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-hydroquinone-bis-(hydroxyethyl)-ether-H-phosphonate (2.5 g) as a white sticky solid having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.9 ( $J_{\text{P}-\text{H}} = 578$  Hz). TLC Rf 0.09 (9:1 dichloromethane/methanol/0.1% triethylamine).

**Example 30****Synthesis of (4,4'-dimethoxytrityl)-N,N-bis-(2-hydroxyethyl)-isonicotinamide-H-phosphonate**

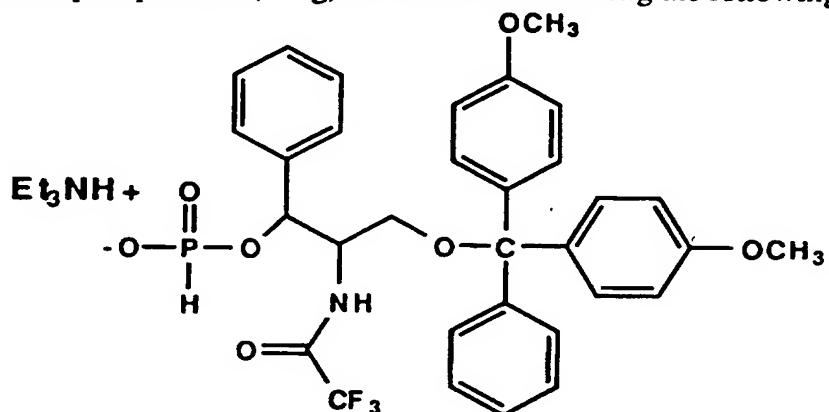
(4,4'-dimethoxytrityl)-N,N-bis-(2-hydroxyethyl)-isonicotinamide (10.2 g, 19.9 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-N,N-bis-(2-hydroxyethyl)-isonicotinamide-H-phosphonate (10.5 g) as a white foam having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.55 (1H, doublet of triplets,  $J_{\text{P}-\text{H}} = 583$  Hz,  $J_{\text{P}-\text{O}-\text{CH}_2} = 7.7$  Hz), 1.98 (1H, doublet of triplets,  $J_{\text{P}-\text{H}} = 583$  Hz,  $J_{\text{P}-\text{O}-\text{CH}_2} = 9.1$  Hz). TLC Rf 0.06 (9:1 dichloromethane/methanol/0.1% triethylamine).

Example 31Synthesis of 3-(4,4'-dimethoxytrityl)-2-amino-1-phenyl-1,3-propanediol-H-phosphonate

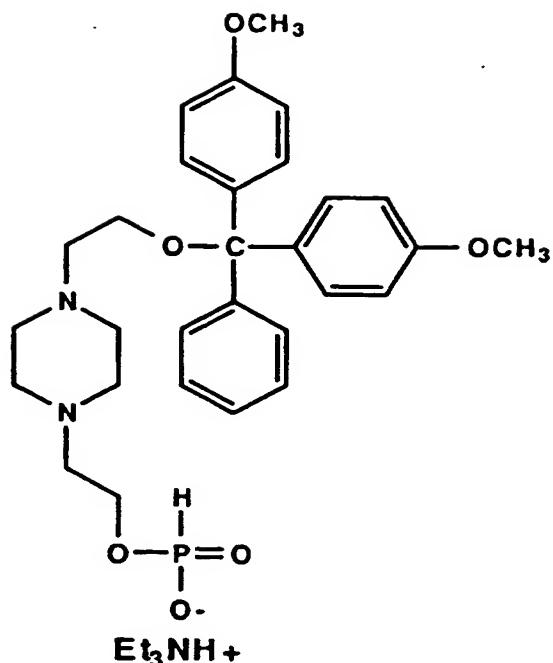
3-(4,4'-dimethoxytrityl)-2-amino-1-phenyl-1,3-propanediol (8.0 g, 14.1 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give 3-(4,4'-dimethoxytrityl)-2-amino-1-phenyl-1,3-propanediol-H-phosphonate (7.1 g) as a white foam having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 0.66 (doublet of doublets,  $J_{\text{P}-\text{H}} = 594$  Hz,  $J_{\text{P}-\text{O}-\text{CH}_2} = 12.1$  Hz). TLC Rf 0.11 (9:1 dichloromethane/methanol/0.1% triethylamine).

Example 32Synthesis of (4,4'-dimethoxytrityl)-1,4-bis-(hydroxyethyl)-piperazine-H-phosphonate

(4,4'-dimethoxytrityl)-1,4-bis-(hydroxyethyl)-piperazine (10.2 g, 21.3 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-1,4-bis-(hydroxyethyl)-piperazine-H-phosphonate (9.1 g) as a white foam having the following structure:

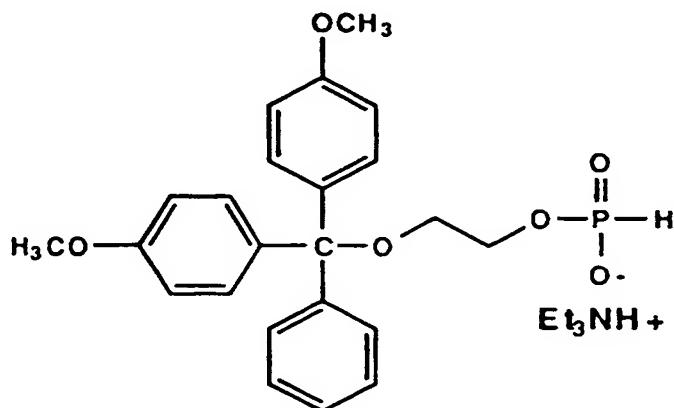


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 4.3 (doublet of triplets,  $J_{\text{P}-\text{H}} = 595$  Hz,  $J_{\text{P}-\text{O}-\text{CH}_2} = 13.7$  Hz). TLC Rf 0.11 (8:2 dichloromethane/methanol/0.1% triethylamine).

### Example 33

#### Synthesis of (4,4'-dimethoxytrityl)-ethylene glycol-H-phosphonate

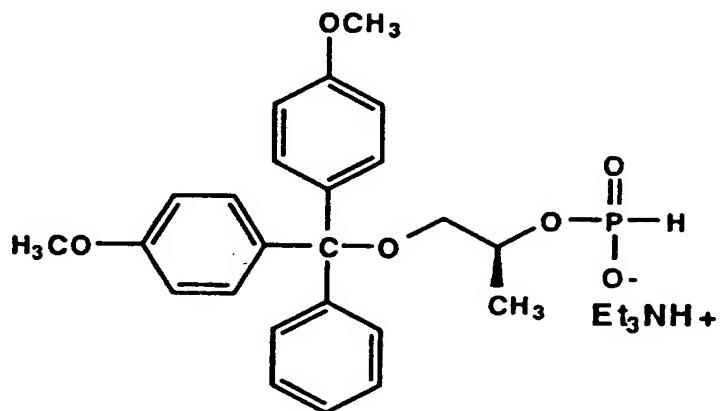
(4,4'-dimethoxytrityl)-ethylene glycol (2.9 g, 7.96 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-ethylene glycol-H-phosphonate (3.8 g) as a yellow oil having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 2.0 ( $J_{\text{P}-\text{H}} = 580$  Hz). TLC Rf 0.09 (95:5 dichloromethane/methanol/0.5% triethylamine).

Example 34Synthesis of 1-(4,4'-dimethoxytrityl)-1-S-1,2-propanediol-H-phosphonate

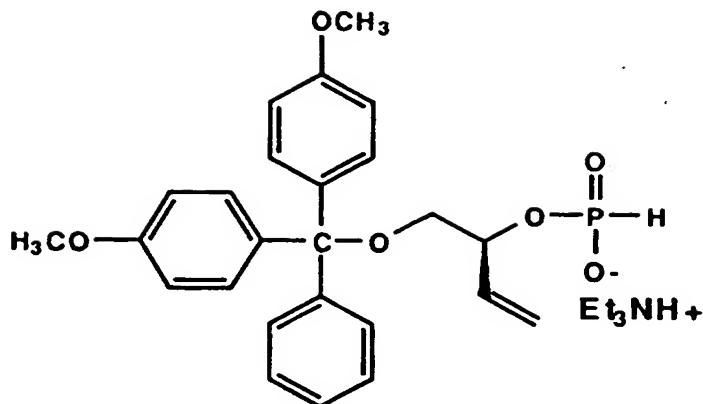
1-(4,4'-dimethoxytrityl)-1,2-propanediol (2.3 g, 6.0 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give 1-(4,4'-dimethoxytrityl)-1,2-propanediol-H-phosphonate (1.75 g) as a yellow gum having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.7 ( $J_{\text{P}-\text{H}} = 616 \text{ Hz}$ ). TLC Rf 0.04 (95:5 dichloromethane/methanol/0.5% triethylamine).

Example 35Synthesis of 1-(4,4'-dimethoxytrityl)-1,2-dihydroxy-3-R-butene-H-phosphonate

1-(4,4'-dimethoxytrityl)-1,2-dihydroxy-3-butene (4.4 g, 11.2 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give 1-(4,4'-dimethoxytrityl)-1,2-dihydroxy-3-butene-H-phosphonate (4.2 g) as a yellow gum having the following structure:

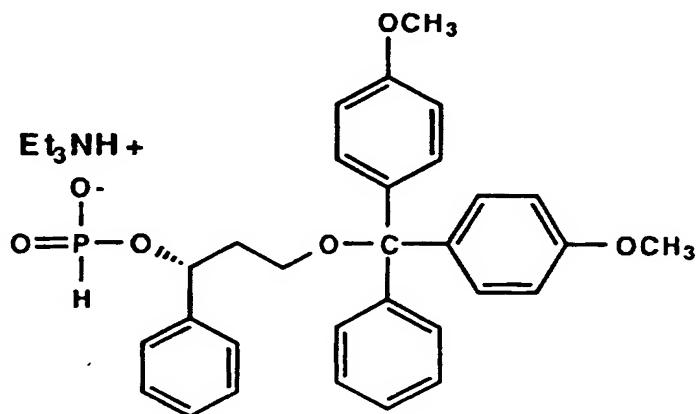


<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 8.17 (J<sub>P-H</sub> = 703 Hz). TLC Rf 0.13 (3:7 ethylacetate/hexane/0.5% triethylamine).

#### Example 36

#### Synthesis of 3-(4,4'-dimethoxytrityl)-1-R-phenyl-1,3-propanediol-H-phosphonate

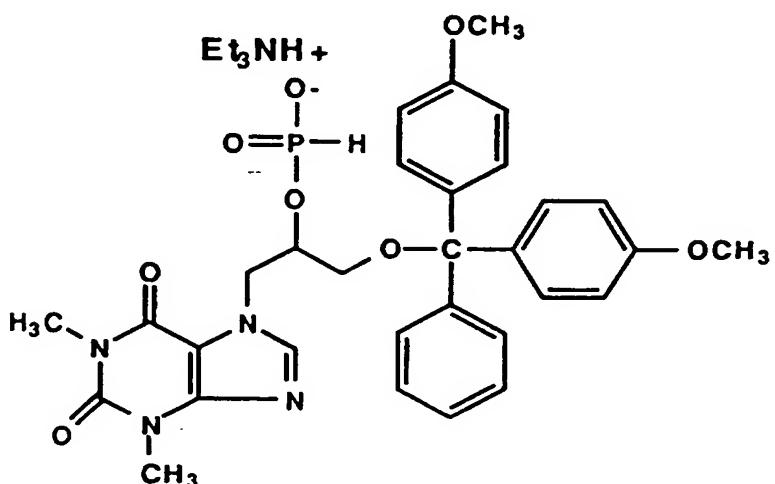
3-(4,4'-dimethoxytrityl)-1-phenyl-1,3-propanediol (2.5 g, 5.5 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give 3-(4,4'-dimethoxytrityl)-1-phenyl-1,3-propanediol-H-phosphonate (2.3 g) as a yellow foam having the following structure:



<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) -1.7 (J<sub>P-H</sub> = 585 Hz). TLC Rf 0.084 (94:6 dichloromethane/methanol/0.5% triethylamine).

Example 37Synthesis of (4,4'-dimethoxytrityl)-2,3-dihydroxypropyl-theophylline-H-phosphonate

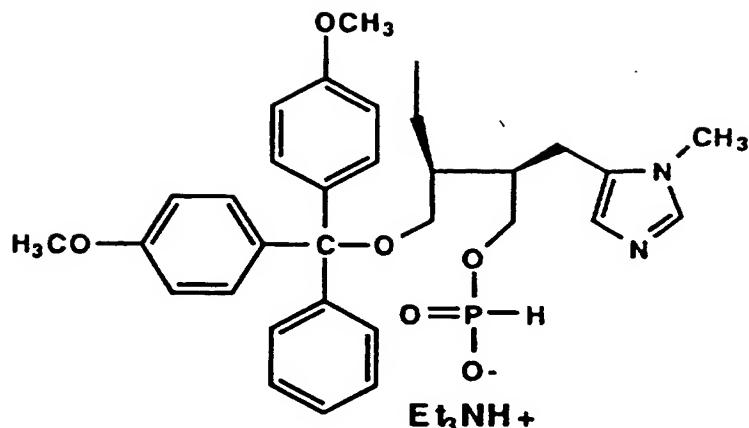
(4,4'-dimethoxytrityl)-2,3-dihydroxypropyl-theophylline (1.9 g, 3.5 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-theophylline-H-phosphonate (2.6 g) as a white foam having the following structure:



<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) 0.76 (doublet of doublets, J<sub>P-H</sub> = 584 Hz, J<sub>P-O-CH<sub>3</sub></sub> = 8.5 Hz)), TLC Rf 0.34 (94:6 dichloromethane/methanol/0.5% triethylamine).

Example 38Synthesis of (4,4'-dimethoxytrityl)-pilocarpine-H-phosphonate

(4,4'-dimethoxytrityl)-pilocarpine (1.2 g, 1.8 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-pilocarpine-H-phosphonate (1.5 g) as a pale yellow foam having the following structure:

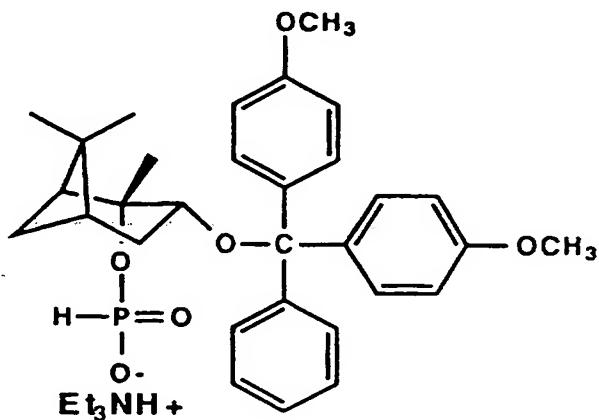


$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.8 ( $J_{\text{P}-\text{H}} = 573$  Hz). TLC Rf 0.122 (95:5 dichloromethane/methanol/1% triethylamine).

### Example 39

#### Synthesis of (4,4'-dimethoxytrityl)-1S,2S,3R,5S-(+)-pinanediol-H-phosphonate

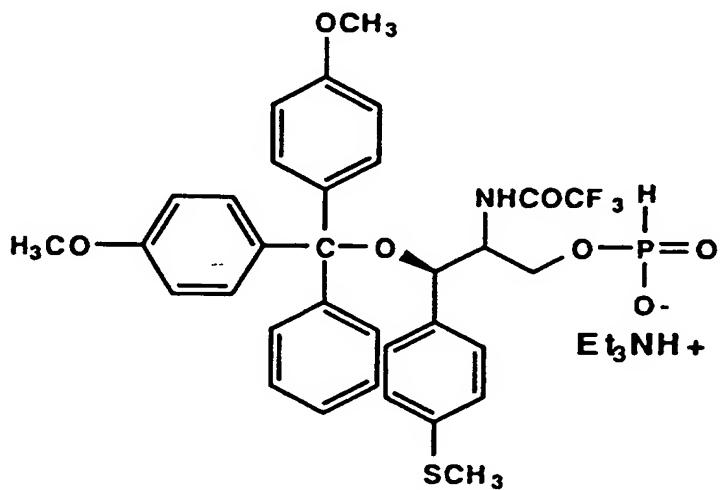
(4,4'-dimethoxytrityl)-1S,2S,3R,5S-(+)-pinanediol (2.51 g, 5.36 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-1S,2S,3R,5S-(+)-pinanediol-H-phosphonate (1.8 g) as a pale yellow foam having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) -1.7 ( $J_{\text{P}-\text{H}} = 585$  Hz). TLC Rf 0.234 (95:5 dichloromethane/methanol/0.5% triethylamine).

Example 40Synthesis of (4,4'-dimethoxytrityl)-thiomicamine-trifluoroacetamide-H-phosphonate

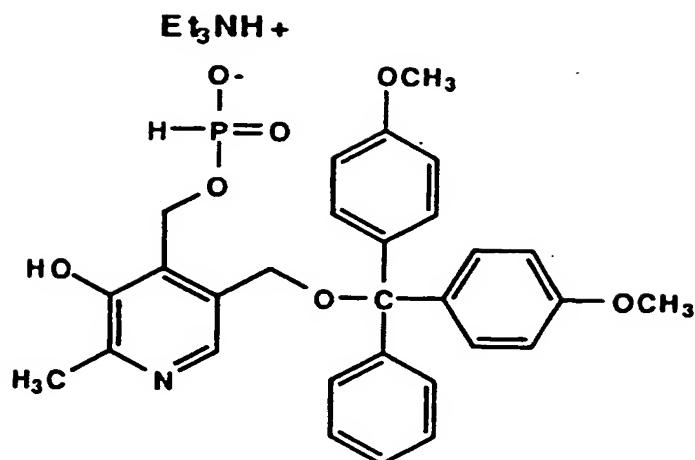
(4,4'-dimethoxytrityl)-thiomicamine-trifluoroacetamide (3.6 g, 6.0 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-thiomicamine-trifluoroacetamide-H-phosphonate (3.0 g) as a pale yellow foam having the following structure:



<sup>31</sup>P NMR (202 MHz) DMSO δ (ppm) -1.7 (J<sub>P-H</sub> = 585 Hz). TLC R<sub>f</sub> 0.234 (95:5 dichloromethane/methanol/0.5% triethylamine).

Example 41Synthesis of (4,4'-dimethoxytrityl)-pyridoxine-H-phosphonate

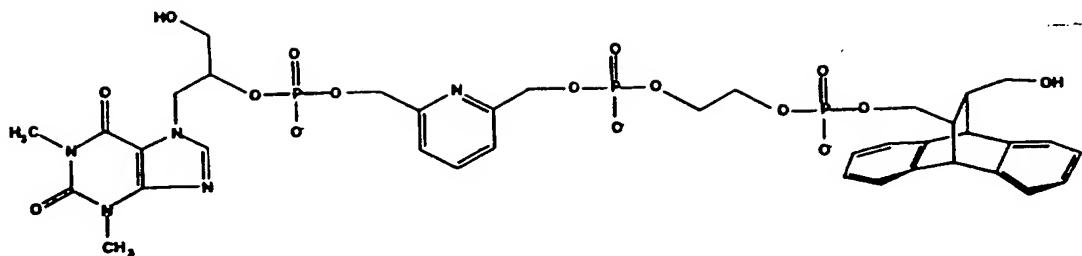
(4,4'-dimethoxytrityl)-pyridoxine (4.57 g, 9.69 mmol), prepared according to the protocol used in example 4, was treated according to the protocol used in example 22 to give (4,4'-dimethoxytrityl)-pyridoxine-H-phosphonate (3.9 g) as a white foam having the following structure:



$^{31}\text{P}$  NMR (202 MHz) DMSO  $\delta$  (ppm) 1.9 ( $J_{\text{P}-\text{H}} = 587$  Hz). TLC Rf 0.33 (8:2 dichloromethane/methanol/0.1% triethylamine).

**Example 42**  
**Synthesis of a Tetrameric Phosphorus Ester**

Samples of the monomers from examples 15, 7, 12 and 5 were separately diluted in anhydrous acetonitrile to give a 0.2 M solution of each. These solutions were used in a standard amidite syntheses on an automated DNA synthesizer using the 1  $\mu\text{mol}$  phosphoramidite cycle with an extended coupling time of 5 min. The oligomer was synthesized on a solid support that had been initially derivatized with the chemical phosphorylation reagent (2-cyanoethoxy)-2-(2'-O-4,4'-dimethoxytrityloxyethylsulfonyl)ethoxy-N,N'-diisopropylaminophosphine. After the addition of each of the monomers to the column, added in the order specified above, the efficiency of the coupling was checked by monitoring of the trityl colors from each cycle, and was determined to be in excess of 95% throughout all of the coupling steps. After the addition of the monomers, deoxycytidine phosphoramidite was added for labelling purposes. The column was then treated with concentrated ammonium hydroxide for 4 h at 55 °C and the ammonia removed by bubbling the solution with nitrogen for 20 min. The solution was lyophilized, and the residue dissolved in water and purified on a C18 HPLC column using a gradient of acetonitrile (solvent B)/triethylammonium acetate, pH 7 (solvent A). Gradient: 8-20% B over 24 min, then 20-40% B over 10 min. The material eluting at 14.75 min was lyophilized to give an oligomer with the following structure:



**Example 43**  
**Synthesis of a Fluorescein-Labelled Pentamer Library**

Samples of the monomers from examples 1-10 were separately diluted in anhydrous acetonitrile to give 2 mL of a 0.2 M solution of each. These solutions were used in a standard amidite syntheses on three Applied Biosystems 394 Automated DNA synthesizers using the 1  $\mu$ mol phosphoramidite cycle with an extended coupling time of 5 min for each. The library was synthesized on ten columns each loaded with 10 mg of 100  $\mu$ m controlled pore glass support that had been derivatized with the chemical phosphorylation reagent (2-cyanoethoxy)-2-(2'-O-4,4'-dimethoxytrityl-oxyethylsulfonyl)ethoxy-N,N'-diisopropyl-aminophosphine. After the addition of each of the monomers to their specified columns the efficiency of the coupling was checked by monitoring of the trityl colors from each synthesis, and was determined to be in excess of 90% throughout all of the coupling steps. After each synthesis cycle, the resins were removed from the columns and pooled and divided using the isopycnic slurry method. The synthesis columns were then weighed to ensure an even division of the resin between the columns, the variation of which was determined to be less than 5% throughout the four pooling and dividing steps. After the addition of the fifth amidite, the oligomers in each of the ten columns were fluoresceinated using fluorescein 6-FAM amidite (Applied Biosystems, Foster City, CA). The resin from each column was then removed, and treated with concentrated ammonium hydroxide for 4 h at 55 °C. The resultant suspensions were filtered and the ammonia removed from the filtrate by bubbling the solutions with nitrogen for 20 min. The ten solutions were then lyophilized and the residues dissolved in water and purified over an OPC cartridge (Applied Biosystems) using the standard procedure as provided by the manufacturer. The products from the purification were then analyzed by electrophoresis on a non-denaturing polyacrylamide gel. Each of the ten products ran as a single band, with almost the same mobility as a pentameric deoxyoligonucleotide. The library was then assembled by mixing all of these solutions and lyophilizing the resultant mixture.

Example 44  
Synthesis of a <sup>32</sup>P-Labelled Trimer Phosphoramidate Library.

Samples of the monomers from examples 23, 27, 29, 32-37 and 39 were separately diluted in 1:1 anhydrous acetonitrile/pyridine to give a 0.12 M solution of each. These solutions were used in two automated DNA synthesizers using a modified H-phosphonate cycle with an extended coupling time of 5 minutes for each monomer. The library was synthesized on 10 columns each loaded with 60 mg of long chain alkylamine controlled pore glass (37-70 µm, 44 umol/g), that had been derivatized with dimethoxytrityl-thymidine. Each column was initially reacted with the chemical phosphorylating reagent (2-cyanoethoxy)-2-(2'-O-4,4'-dimethoxytrityloxyethylsulfonyl)ethoxy-N, N'-diisopropylaminophosphine using the standard phosphoramidite procedure. Each of the 10 monomers was then coupled to its specified column, monomer one to column one, monomer two to column two, etc., up to monomer 10 to column 10. The efficiency of each coupling was checked by monitoring the trityl colors from each synthesis, and was determined to be in excess of 90% throughout all of the coupling steps. After addition of the 10 monomers, the resins were then removed from the columns and pooled and divided using the isopycnic slurry method. The synthesis columns were weighed to ensure an even division of the resin among the columns. After a second addition of all 10 monomers as previously described, followed by a second pool-and-divide procedure, the non-nucleotide H-phosphonate linkages were converted to phosphoramidate linkages using the following oxidation procedure with each column being oxidized with one of the amines listed below:

1. 2-(2-Aminoethyl)pyridine,
2. 1-(2-Aminoethyl)piperidine,
3. 2-(3, 4-Dimethoxyphenyl)ethylamine,
4. 4-(2-Aminoethyl)-morpholine,
5. 1-(3-Aminopropyl)-2-pyrrolidinone,
6. 2-(3-chlorophenyl)ethylamine,
7. 4-Bromophenethylamine,
8. N-(3-trifluoromethyl)phenylpiperazine,
9. 3,3-Diphenylpropylamine,
10. Thiomorpholine,
11. 1-(2-Pyridyl)piperazine,
12. Hexamethyleneimine,
13. cis-2,6-Dimethylmorpholine,
14. 1-(4-fluorophenyl) piperazine,

## 15. 4-Fluorophenethylamine,

## 16. 3,5-Dimethylpiperidine.

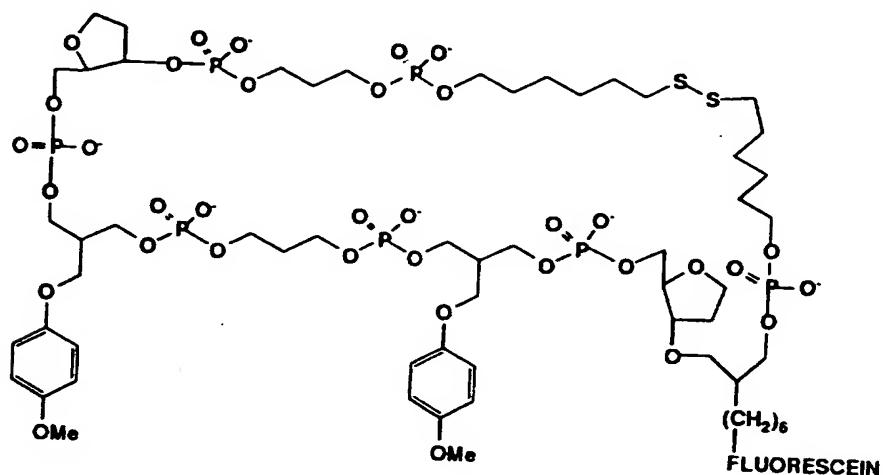
A 10 % solution of each amine was prepared individually in carbon tetrachloride and the H-phosphonate linkages of the oligomers were then oxidized to phosphoramidates using the freshly prepared solution of the required amine for 14 minutes. The amine solutions were then flushed from the columns and the supports were washed extensively with acetonitrile. The supports were removed from the columns and pooled and divided as described above. A second round of monomer addition followed by pooling, dividing and amine addition was carried out, followed by addition of thymidine-H-phosphonate at the last step to provide an attachment site for the  $^{32}\text{P}$  label. The supports were then removed from the columns and treated with concentrated ammonium hydroxide for 5.5 hours at 55°C. The resultant suspensions were filtered and the ammonia removed from the filtrates by bubbling the solutions with nitrogen for 20 minutes. The sixteen solutions then were lyophilized, redissolved in water and purified by reversed-phase HPLC using a gradient of 5-100% acetonitrile in TEAA, 0.1 M, pH 7. The products from these purifications were combined and resultant mixture was lyophilized to give the oligomeric phosphoramidate library which was then labelled with  $^{32}\text{P}$  by a conventional method using T4 polynucleotide kinase.

Example 45Synthesis of a Disulfide-Bridged Phosphorus Ester Oligomer

The 3',5'-dithiol oligomer was synthesized on an Applied Biosystems Model 394 DNA synthesizer on a 1  $\mu\text{mol}$  scale using 0.02 M solution of iodine for all of the oxidation steps. 1-O-Dimethoxytrityl-hexyl-disulfide-1'-(2-cyanoethyl)-N,N-diisopropyl-phosphoramidite (thiol modifier C6 S-S, Glen Research, Stirling, VA) was directly coupled to a nucleoside residue anchored to a controlled pore glass solid support (i.e. a thymidine column) and the coupling time was extended to 15 min in this cycle and the following two phosphoramidite coupling cycles. The monomer phosphoramidites as specified in examples 2, 1, 3, 2, 3 and 1 were then incorporated sequentially in the exact order specified using a coupling time of 5 min for each. Fluorescein phosphoramidite (ClonTech Co.) was then added followed by the thiol modifier C6 S-S using a 15 min coupling time. Coupling efficiencies for the monomers averaged 88%, as-determined by trityl assay.

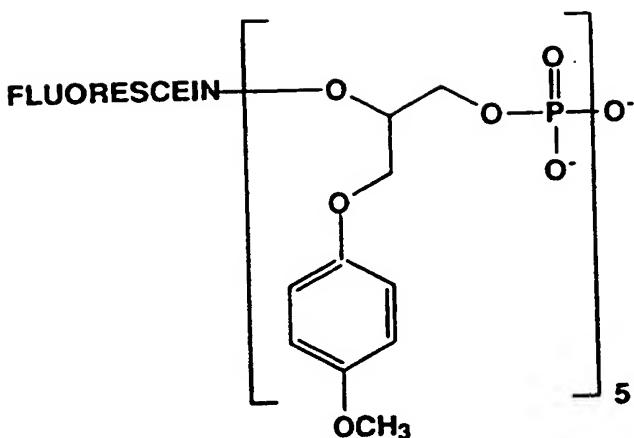
After the synthesis and cleavage from the solid support, the solution was concentrated to a small volume and the tritylated oligomer was isolated by HPLC on a preparative C18 column using a gradient of acetonitrile in 0.1 M triethyl-ammonium acetate (TEAA) buffer (5-30% over 15 min then 30-55% over 30 min, then at 55% for 20 min). Fractions eluting at 47-51 min were evaporated to dryness and redissolved in 0.1 M

TEAA buffer (pH 8.0, 0.5 mL). After addition of dithiothreitol (DTT, 10 mg), the resulting mixture was stirred under an argon blanket at room temperature for 4 hr to cleave the disulfide bonds and liberate free thiol groups at both ends of the oligomer. The disulfide bond cleavage reaction was monitored by HPLC using a C<sub>18</sub> analytical column. The byproducts were removed by extraction with ethyl acetate (3 X) and ether (2 X) and the intermediate oligomer dithiol was purified by HPLC on a preparative C<sub>18</sub> column using a gradient of 2-20% acetonitrile in 0.1 M TEAA buffer over 15 min followed by 20-50% over 30 min, then 50-65% over 15 min. The fraction eluting between 30-35 min was evaporated to dryness, dissolved in triethylammonium acetate buffer pH 8.0 (1 mL) and oxidized with oxygen with stirring at room temperature for 1 day. The reaction mixture was analyzed by HPLC on a C<sub>18</sub> analytical column using a gradient of acetonitrile in 0.1 M TEAA buffer (5-30% over 10 min, then 15-50% over 23 min, then 50-80% over 7 min. The appropriate fractions were lyophilized to give a disulfide-bridged oligomer (HPLC retention time 24.0 min) with the following structure:



Example 46  
Serum Stability of a Non-nucleotide Oligomeric Phosphorus Ester

The H-phosphonate synthesis method was used to prepare a fluorescein-labelled pentamer with the following structure:



Portions (9.6 nmol) of this material in separate tubes were each dissolved in water (105 µL), added to human serum (500 µL) and incubated at 37°. At various time points, samples were filtered through a 0.45 µm filter and analyzed by ion exchange HPLC using a Dionex PA-100 column (4 x 250 mm). The column was eluted with a gradient of 25 mM Tris chloride, 1M ammonium chloride, pH 8 (buffer B) in 25 mM Tris chloride, pH 8 (buffer A), using 15-65% B over 19 min, then 65% B for 2 min, then 65-75% B over 2 min. The area under the peak corresponding to starting material was measured versus time to determine the degradation rate. By this method the  $t_{1/2}$  was determined to be greater than 1 month. A 20-base oligodeoxynucleotide was used as a control, and under the same conditions, the  $t_{1/2}$  for this oligonucleotide was determined to be 7 min.

#### Example 47

#### Binding of a Library of Phosphorus Ester Oligomers to Protein Targets

The oligomeric library of Example 40 was screened against the target proteins IL-4 and IFN $\gamma$  for high affinity ligands, using the COMPILE method as outlined in copending patent application "Determination and Identification of Active Compounds in a Compound Library," U. S. patent application serial number 08/223,519. Binding reactions consisting of 2.5 µM oligomeric library and 1 µM target protein as well as a protein-minus control (data not shown) were prepared in a buffer containing 150 mM NaCl, 3 mM MgCl<sub>2</sub> and 25 mM Tris-HCl (pH 7.5). The reactions were equilibrated for 30 min at room temperature prior to centrifugation through Microcon-10 spin filtration units (M.Wt.cutoff = 10,000). The M.Wt. cutoff of the filtration unit was selected to allow free passage of the oligomeric library members but not the target protein or the oligomeric molecules bound to it.

The centrifugation was adjusted to allow approximately 270  $\mu\text{L}$  of the total reaction volume of 400  $\mu\text{L}$  to flow through the membrane, resulting in the loss of approximately 67% of the unbound oligomeric library from the retained sample. The volume of the retained sample was brought up to the original 400  $\mu\text{L}$  with buffer (containing 20% of the original target concentration to compensate for losses/inactivation of the target during the experiment) and the selection was repeated.

The results of this serial selection are presented in Figure 1. The concentration of the oligomeric library present in the retained fraction (as measured by fluorescence of the fluorescein tag incorporated into the library members) is plotted as a function of the rounds of selection - i.e. the number of dialysis-based separations. It should be noted here that any method of separation of target-bound library members from unbound library members, even separation not based on size, could have been chosen. Data are not presented for the first two rounds of selection since the library was in excess over target at that point, and we would not observe protein dependent retention of the library.

As shown in Figure 1, the oligomeric library members are lost at approximately the same rate for both IL-4 and IFN $\gamma$  (and for the protein-minus control - data not shown) over the first seven rounds of selection. At this point, however, the IFN $\gamma$  reaction begins to retain a progressively larger fraction of the library whereas the IL-4 reaction fails to show such increased proportion of retention.

The progressive increase in the fraction of compounds retained at each round, as observed with the IFN $\gamma$  reaction, is evidence of a subpopulation of the original library that binds to the target. This subpopulation is enriched with each successive round of selection due to selective retention of compounds in the library that bind to the target protein. From the target protein concentration and the curve shown in Figure 1, it is possible to calculate that this subpopulation of IFN $\gamma$ -binding oligomeric species is about 1% of the original mixture, and that it binds with Kd of approximately 250 nM.

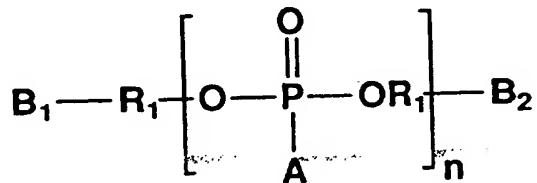
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What Is Claimed Is:

1. A phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, alkyl- or aryl-substituted amino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonooester;

R<sub>1</sub> can be the same or different in each monomeric unit, and in at least one of the non-nucleotide monomeric units, R<sub>1</sub> is independently selected from the group consisting of a

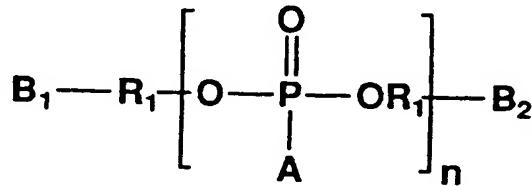
condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring-substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label; and  
n is at least one.

2. The oligomer of claim 1 wherein the hydrogen bond donor contains a hydroxyl, amide, imide or thiol group.
3. The oligomer of claim 1 wherein the hydrogen bond donor is a basic compound.
4. The oligomer of claim 3 wherein the basic compound contains an amine moiety.
5. The oligomer of claim 3 wherein the basic compound is a substituted or unsubstituted thiazole.
6. The oligomer of claim 1 wherein R<sub>1</sub> is a hydrogen bond acceptor which further includes an amine or ether moiety.
7. The oligomer of claim 1 wherein the hydrogen bond acceptor is selected from a 5-substituted 2-hydroxymethyl-3-hydroxy-tetrahydrofuran and a bis(hydroxyalkyl)-substituted heterocycle.
8. The oligomer of claim 1 wherein the hydrogen bond acceptor is a substituted or unsubstituted theophylline.
9. The oligomer of claim 1 wherein R<sub>1</sub> is a hydrophobic functionality selected from the group consisting of aromatic rings, alkanes, cycloalkanes and aromatic rings fused to cycloalkanes.
10. The oligomer of claim 9 wherein R<sub>1</sub> is a hydrophobic functionality selected from a substituted alkane, a 3,3-disubstituted 3-amino-1,2-propanediol, a substituted or

unsubstituted alicyclic ring wherein the ring size is from 4-12, a 3-substituted indole, a substituted or unsubstituted hydroxyalkyl phenol and an alicyclic dicarboxylic acid.

11. The oligomer of claim 1 wherein the anionic functionality is an alicyclic dicarboxylic acid moiety.
12. The oligomer of claim 11 wherein the anionic functionality is a tricyclononene dicarboxylic acid moiety.
13. The oligomer of claim 1 wherein the anionic functionality is a cyclopentane acetic acid moiety.
14. The oligomer of claim 1 wherein the cationic functionality is a bis(hydroxyalkyl)-substituted nitrogen-containing heterocycle.
15. The oligomer of claim 14 wherein the cationic functionality is selected from a substituted alkane and a 3,3-disubstituted 3-amino-1,2-propanediol.
16. The oligomer of claim 1 wherein in at least one of the monomeric units R<sub>1</sub> is a heterocyclic, an alicyclic or a polycyclic ring system.
17. The oligomer of claim 16 wherein the heterocyclic ring system contains an indole, thiazole, imidazole, purine or pyrimidine ring.
18. The oligomer of claim 16 wherein the alicyclic ring system contains a cyclopentane or cyclooctane ring.
19. The oligomer of claim 18 wherein the cyclopentane or cyclooctane ring system is substituted with at least one carboxylic acid moiety.
20. The oligomer of claim 16 wherein the polycyclic ring system contains a bicyclic or tricyclic ring or is a polycyclic arene.
21. The oligomer of claim 20 wherein the bicyclic ring is a bicyclic alkane.
22. The oligomer of claim 21 wherein the bicyclic ring is bicycloheptane.
23. The oligomer of claim 20 wherein the tricyclic ring is an alkene.

24. The oligomer of claim 23 wherein the alkene is tricyclononene.
25. The oligomer of claim 20 wherein the polycyclic arene is diphenyl bicyclooctane.
26. The oligomer of claim 1 wherein A is selected from the group consisting of 2-(2-aminoethyl)pyridine, 1-(2-aminoethyl)piperidine, 2-(3, 4-dimethoxy-phenyl)ethylamine, 4-(2-aminoethyl)-morpholine, 1-(3-aminopropyl)-4-methylpiperazine, 1-(3-aminopropyl)-2-pyrrolidinone, 1-(3-aminopropyl)imidazole, 1-(2-aminoethyl)pyrrolidine, 3-aminopropionitrile, 2-(2-aminoethyl)-1-methyl-pyrrolidine, 4-fluorophenethylamine, 4-bromophenethylamine, aminomethyl-cyclopropane, 3,3-diphenylpropylamine, formylpiperazine, trifluoromethyl-phenylpiperazine, thiomorpholine, 1-(2-pyridyl)piperazine, homopiperazine, hexamethyleneimine, cis-2,6-dimethylmorpholine, 2,5-dimethylphenylpiperazine, 3,5-dimethylpiperidine, 1-(4-fluorophenyl) piperazine, N-(3, 4-dichlorophenyl)piperazine, 2-(4-chlorophenyl)ethylamine, 4-piperazineacetophenone, 4-piperidinopiperidine, 2-thiophenemethylamine, furfurylamine, heptamethyleneimine, and 1-(4-methoxyphenyl)piperazine.
27. The oligomer of claim 1 wherein A is an N-linked amino acid.
28. A phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, alkyl- or aryl-substituted amino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonoester;

R<sub>1</sub> can be the same or different in each monomeric unit and in at least one of the monomeric units is independently selected from a condensation product of:

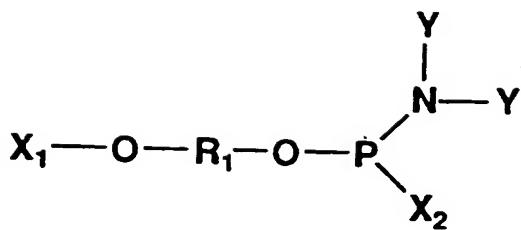
- (i) an aliphatic acyclic diol wherein the diol hydroxyl groups are non-vicinal or are substituted;
- (ii) a purine- or pyrimidine-substituted variant of the diols of (i) or of aliphatic acyclic vicinal diols;
- (iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;
- (iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;
- (v) a hydroxy- or hydroxyalkyl-substituted tetrahydrofuran;
- (vi) an indole-substituted acyclic aliphatic diol;
- (vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;
- (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
- (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline, any of which can further include a detectable label; and  
n is at least one.

29. The oligomer of claim 28 wherein n is 2-20.

30. The oligomer of claim 28 wherein A is selected from the group consisting of 2-(2-aminoethyl)pyridine, 1-(2-aminoethyl)piperidine, 2-(3, 4-dimethoxyphenyl)ethylamine, 4-(2-aminoethyl)-morpholine, 1-(3-aminopropyl)-4-methylpiperazine, 1-(3-aminopropyl)-2-pyrrolidinone, 1-(3-aminopropyl)imidazole, 1-(2-aminoethyl)pyrrolidine, 3-aminopropionitrile, 2-(2-aminoethyl)-1-methyl-pyrrolidine, 4-fluorophenethylamine, 4-bromophenethylamine, aminomethyl-cyclopropane, 3,3-diphenylpropylamine, formylpiperazine, trifluoromethylphenylpiperazine, thiomorpholine, 1-(2-pyridyl)piperazine, homopiperazine, hexamethyleneimine, cis-2,6-dimethylmorpholine, 2,5-dimethylphenylpiperazine, 3,5-dimethylpiperidine, 1-(4-fluorophenyl) piperazine, N-(3, 4-dichlorophenyl)piperazine, 2-(4-chlorophenyl)ethylamine, 4-piperazineacetophenone, 4-piperidinopiperidine, 2-thiophenemethylamine, furfurylamine, heptamethyleneimine and 1-(4-methoxyphenyl)piperazine.

31. The oligomer of claim 28 wherein A is an N-linked amino acid.

32. A non-nucleotide monomeric unit having the structure:



wherein

$X_1$  is a protecting group;

$X_2$  is a branched or unbranched lower alkyl group or a substituted or unsubstituted alkoxy group;

$Y$  is a branched or unbranched lower alkyl group; and

$R_1$  is selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label.

33. The non-nucleotide monomeric unit of claim 32 wherein the anionic functionality is a cyclopentane acetic acid moiety.

34. The non-nucleotide monomeric unit of claim 32 wherein the cationic functionality is a bis(hydroxyalkyl)-substituted nitrogen-containing heterocycle.

35. The non-nucleotide monomeric unit of claim 32 wherein the cationic functionality is a 3,3-disubstituted 3-amino-1,2-propanediol.

36. The non-nucleotide monomeric unit of claim 32 wherein  $R_1$  is a condensation product of:

(i) an aliphatic acyclic diol wherein the diol hydroxyl groups are non-vicinal or are substituted;

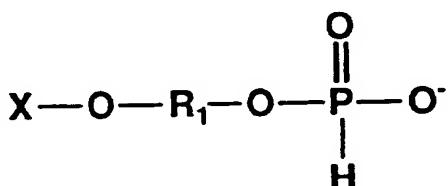
(ii) a purine or pyrimidine substituted variant of the diols of (i) or of aliphatic acyclic hydrocarbon vicinal diols;

(iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;

(iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;

- (v) a hydroxy or hydroxyalkyl substituted tetrahydrofuran;
- (vi) an indole substituted acyclic aliphatic diol;
- (vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;
- (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
- (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline, and in which R<sub>1</sub> can optionally be substituted with a detectable label.

37. A non-nucleotide monomeric unit having the structure:



wherein

X is a protecting group;  
R<sub>1</sub> is selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label.

38. The non-nucleotide monomeric unit of claim 37 wherein the anionic functionality is a cyclopentane acetic acid moiety.

39. The non-nucleotide monomeric unit of claim 37 wherein the cationic functionality is a bis(hydroxyalkyl)-substituted nitrogen-containing heterocycle.

40. The non-nucleotide monomeric unit of claim 37 wherein the cationic functionality is a 3,3-disubstituted 3-amino-1,2-propanediol.

41. The non-nucleotide monomeric unit of claim 37 wherein R<sub>1</sub> is a condensation product of:

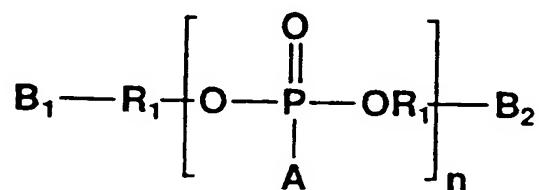
- (i) an aliphatic acyclic hydrocarbon diol wherein the diol hydroxyl groups are non-vicinal or are substituted;
- (ii) a purine- or pyrimidine-substituted variant of the diols of (i) or of aliphatic acyclic vicinal diols;
- (iii) an acyclic aliphatic diol having an amino group with at least one hydrogen substitution moiety;
- (iv) an alicyclic or polycyclic diol, optionally substituted with a carboxy or carboxyalkyl substituent;
- (v) a hydroxy- or hydroxyalkyl-substituted tetrahydrofuran;
- (vi) an indole-substituted acyclic aliphatic diol;
- (vii) an aromatic ring or ring system having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl;
- (viii) a heterocyclic compound having two substitutions independently selected from the group consisting of hydroxy or hydroxyalkyl; and
- (ix) a dihydroxyalkyl thiazole or dihydroxyalkyl oxazoline, and in which R<sub>1</sub> can optionally be substituted with a detectable label.

**42.** A combinatorial library comprising a mixture of the oligomers of claim 1.

**43.** A combinatorial library comprising a mixture of the oligomers of claim 16.

**44.** A combinatorial library comprising a mixture of the oligomers of claim 28.

**45.** A phosphorus ester oligomer of monomeric units, which oligomer has the structure:



wherein

A can be the same or different in each monomeric unit and each is independently selected from the group consisting of oxygen, sulfur, lower alkyl, alkyl- or aryl-substituted amino and aminoalkyl;

B<sub>1</sub> and B<sub>2</sub> can be the same or different and each is independently selected from hydrogen, lower alkyl, a labelling group, a protecting group, a phosphoramidate or a phosphomonoester;

R<sub>1</sub> can be the same or different in each monomeric unit, and is a nucleoside moiety or, in at least one of the non-nucleotide monomeric units, R<sub>1</sub> is independently selected from the group consisting of a condensation product of (i) a non-vicinal diol attached to a hydrogen bond donor functionality; (ii) a hydrogen bond acceptor selected from an ether, a purine or pyrimidine substituted 1,2-diol or a disubstituted heterocycle; (iii) a non-vicinal diol attached to a hydrophobic functionality or a vicinal diol attached to an aliphatic or alicyclic hydrophobic functionality (iv) a diol attached to a ring substituted anionic functionality and (v) a cationic moiety attached to a non-vicinal or alicyclic diol, any of which can further include a detectable label; and

n is at least one.

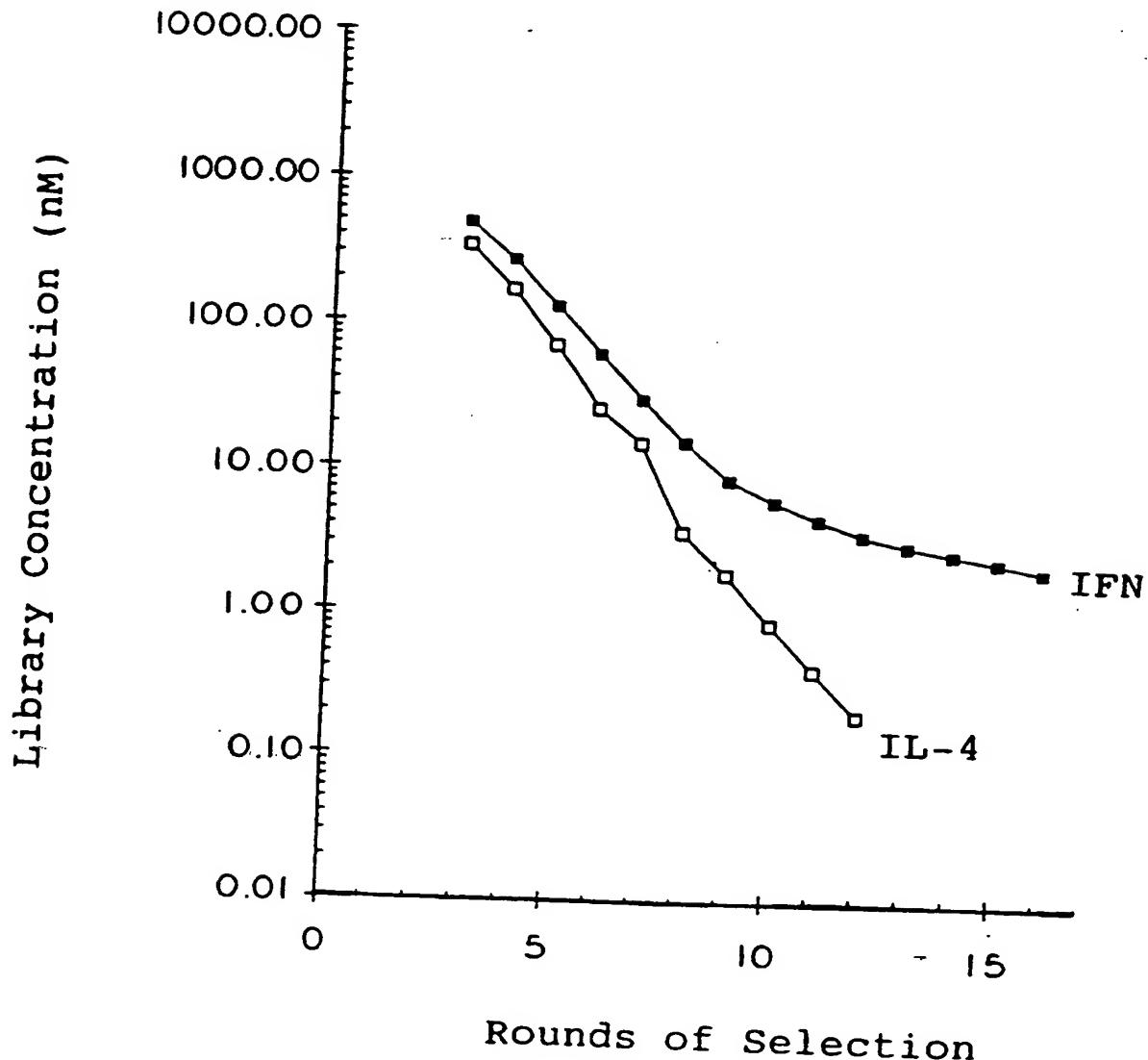
46. The oligomer of claim 45 wherein the nucleoside moiety is a ribonucleoside, 2'-O-methyl-ribonucleoside or a deoxyribonucleoside moiety.

47. The oligomer of claim 1 wherein n is 2-20.

1 / 1

F I G. 1

Oligomeric Library Selection against  
IL-4 and IFN $\gamma$  targets



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## INTERNATIONAL SEARCH REPORT

International application No.

PCT/US96/00369

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) :C07F 9/09, 9/117, 9/12

US CL :546/22; 548/119; 549/220

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 546/22; 548/119; 549/220

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

APS - phosphorus ester oligomers

CAS structure search in U.S. parent application

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US, A, 3,891,727 (WEIL) 24 JUNE 1975, See Column 6, lines 55-65, Columns 7-12 and Examples.	1, 9-10, 28, 42, 44-45
X	US, A, 3,772,340 (SHAMRAO ET AL.) 13 NOVEMBER 1973, See Column 4, lines 10-45.	1, 16, 20, 28, 42-45
X	US, A, 5,364,564 (KOGA) 15 NOVEMBER 1994, See Column 1, lines 11-35.	1, 9-10, 16, 20, 28, 42-45.
X	US, A, 5,153,347 (LLOYD) 06 OCTOBER 1992, See Column 1, lines 15-30 and Examples.	1-4, 6, 9-10, 28, 42, 44-45.

 Further documents are listed in the continuation of Box C.  See patent family annex.

* Special categories of cited documents:	"T"	later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A* document defining the general state of the art which is not considered to be of particular relevance	"X"	document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*E* earlier document published on or after the international filing date	"Y"	document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"&"	document/member of the same patent family
*O* document referring to an oral disclosure, use, exhibition or other means		
*P* document published prior to the international filing date but later than the priority date claimed		

Date of the actual completion of the international search

03 MAY 1996

Date of mailing of the international search report

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